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3 **Miniaturized and modified QuEChERS method with mesostructured**
4 **silica as clean-up sorbent for pyrrolizidine alkaloids determination in**
5 **aromatic herbs**

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17 **ABSTRACT**

18 This work proposes the miniaturization and modification of the QuEChERS strategy using different
19 large pore mesostructured silicas, non-modified and modified with amino groups (NH₂), as dispersive
20 clean-up sorbents for multi-component extraction of 21 pyrrolizidine alkaloids from different aromatic
21 herbs, combined with ultra-high performance liquid chromatography coupled to tandem mass
22 spectrometry analysis. The procedure was miniaturized by reducing the amounts of sample (0.2 g),
23 solvents (2 mL), clean-up sorbents (25 mg sorbent + 150 mg MgSO₄) and partitioning salts (0.65 g)
24 employed. Best results were achieved using mesostructured silicas (LP-MS-NH₂) than conventional
25 PSA. The method was validated (overall recoveries 73-105%) and applied to the analysis of 17 samples.
26 All the samples were contaminated with PAs (average concentration 262 µg/Kg). Thyme and basil
27 samples were the most contaminated, whereas rosemary was the least. Lasiocarpine, senecivernine N-
28 oxide and europine N-oxide were the main PAs that contributed to their contamination.

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30 **Keywords:** pyrrolizidine alkaloids; aromatic herbs; mesostructured silicas; µ-QuEChERS; UHPLC-
31 MS/MS; food safety

32 1. Introduction

33 Pyrrolizidine alkaloids (PAs) and their oxidized forms (pyrrolizidine alkaloids *N*-oxides, PANOs)
34 are natural plant toxins, which can be found as potential contaminants in food. Their intake is mainly
35 associated with liver damage, among other health issues (Dusemund et al., 2018). The main sources of
36 PAs consumption in humans are plant-derived products contaminated with these alkaloids due to the
37 accidental inclusion of weeds or impurities from PAs-producing plants during harvest. Nonetheless, the
38 horizontal natural transfer of PAs/PANOs through the soil among living plants growing nearby or from
39 dead plant materials has also been demonstrated as an alternative contamination path (Selmar et al.,
40 2015; Nowak et al., 2016; Selmar et al., 2019). In this sense, the European Food Safety Authority (EFSA)
41 considers honey, teas, herbal teas and plant-derived food supplements as the main food products likely
42 to be contaminated with PAs and PANOs (EFSA, 2017). However, recent food alerts have notified
43 concerning high levels of these alkaloids in other matrices, such as spices and aromatic herbs,
44 highlighting their occurrence as an important food safety issue that needs to be addressed urgently
45 (RASFF, 2021). Nevertheless, to date, these matrices have gone mostly unnoticed, so works focusing
46 on the detection of these contaminants in these food items are scarce in the literature (Cramer et al.,
47 2013; Kapp, 2017; Picron et al., 2018a; Izcara et al., 2020; Kaltner et al., 2020). Thus, sensitive analytical
48 methods need to be developed to accurately monitor the presence of these compounds in aromatic herbs
49 and spices and ensure food safety. In this sense, due to their potential toxicity and their frequent
50 occurrence, a regulation has recently been published to monitor the occurrence of PAs/PANOs in certain
51 foodstuffs (COMMISSION REGULATION (EU) 2020/2040). In this regulation, maximum
52 concentrations levels have been established for a total of 21 PAs/PANOs (intermedine, lycopsamine,
53 senecionine, senecivermine, seneciphylline, retrorsine, echimidine, lasiocarpine, europine and heliotrine
54 their corresponding *N*-oxides and senkirkine) (Figure 1) in some food products. As well, 14 additional
55 PAs (known to co-elute with one or more of the above 21 compounds) can be also contemplated in these
56 maximum concentration levels whenever the chromatographic method employed is able to individually

57 and separately identified them from the others (COMMISSION REGULATION (EU) 2020/2040). In
58 fact, the coelution of isomers is the main challenge in the analysis of these contaminants. Moreover, the
59 multiresidue determination of these natural contaminants in food samples is a difficult task, as they are
60 subjected to multiple matrix interferences that hinder their extraction and detection because of the high
61 complexity of food samples. Accordingly, a suitable clean-up procedure of the sample or sample extract
62 before its instrumental analysis is crucial to achieve sensitive results and good analytical performance.
63 In this context, the QuEChERS (quick, easy, cheap, effective, rugged and safe) procedure is an
64 appropriate approach, as it involves simultaneous extraction and clean-up of samples and it is designed
65 for the determination of multiple analytes at the same time (Anastassiades et al., 2003). In the original
66 QuEChERS strategy, primary secondary amine (PSA) was used as dispersive clean-up sorbent, as it is
67 useful to remove polar organic acids, polar pigments, some sugars and fatty acids due to its weak anion
68 exchange properties. However, PSA is sometimes not capable of removing excessive interferences in
69 complex matrices (Oellig & Schmid, 2019). For this reason, over the years, the QuEChERS method has
70 been modified by the introduction of other clean-up sorbents, mainly, graphitized carbon black (GCB)
71 and octadecylsilane (C18), which are usually used in combination with PSA (Bruzzoniti et al., 2014;
72 Lawal et al., 2018). This has led researchers to search and evaluate other novel clean-up sorbents for
73 QuEChERS. Consequently, multiwalled carbon nanotubes (MWCNTs) (Zhao et al., 2012; Han et al.,
74 2015; Uclés et al., 2015), magnetic nanoparticles (Li et al., 2014; Zheng et al., 2015), zirconia-based
75 sorbents (Uclés et al., 2015; Urban & Lesueur, 2017), sol-gel organic-inorganic hybrid sorbents (Omar,
76 Irahim and Elbashir, 2014) and an organic polyamine polymer (Oellig & Schmid, 2019) have been
77 proposed as alternative clean-up sorbents for QuEChERS. In this context, ordered mesostructured silicas
78 are sol-gel materials with advanced textural properties (including high surface area, large pore volume,
79 controllable particle size and morphology, well defined pore-size distribution, controllable wall
80 composition, stable aqueous dispersion and excellent chemical, thermal and mechanical stability),
81 making them suitable sorbents for sample preparation (Casado et al., 2017). Moreover, their surface can

82 be easily modified with a wide variety of ligands that can tailor their physical and chemical properties
83 to specific applications. Thus, they can be designed to display different desirable characteristics in
84 adsorption processes. Therefore, according to these advantageous properties, ordered mesostructured
85 silicas could also be used as clean-up sorbents to isolate undesirable matrix interferences and enhance
86 the sensitivity of analytical methods.

87 On the other hand, an important current trend in the analytical field is the development of
88 environmentally friendly methodologies that comply with the Green Analytical Chemistry (GAC)
89 principles, mainly involving a minimum consumption of solvents and samples. This can be achieved by
90 the miniaturization of conventional analytical operations (Casado et al., 2020). In this context, the
91 original QuEChERS procedure has been successfully miniaturized (μ -QuEChERS) and applied in
92 different food matrices (Porto-Figueira, Camacho and Câmara, 2015; Casado et al., 2018; Izcara et al.,
93 2020), leading to cost-effective and environmentally friendly methods. Accordingly, this work, proposes
94 the miniaturization and modification of the original QuEChERS procedure by significantly reducing the
95 amounts of sample, organic solvents, clean-up sorbents and partitioning salts required, and using
96 different ordered mesostructured silicas as dispersive clean-up sorbents for the multi-component
97 extraction of 21 PAs/PANOs from different aromatic herbs (rosemary, basil, thyme and herbs de
98 Provence). To the best of our knowledge, this is the first time that ordered mesostructured silicas are
99 used as dispersive clean-up sorbents in a miniaturized QuEChERS procedure for the determination of
100 PAs and PANOs in food samples or other matrices.

101 **2. Materials and methods**

102 *2.1. Chemicals, reagents and standard solutions*

103 Poly(ethylene glycol)-block-poly(propyleneglycol)-block-poly(ethylene glycol) (EO20PO70EO20,
104 Pluronic 123, M_{av} = 5800 g/mol, d = 1.019 g/mL), tetraethylorthosilicate (TEOS) 98% (M = 208.33
105 g/mol, d = 0.934 g/mL), decane (M = 142.28 g/mol, d = 0.73 g/mL) and (3-aminopropyl)triethoxysilane

106 (M = 221.37 g/mol) were obtained from Sigma-Aldrich (St. Louis, MO, USA). HCl 37% (M = 36.46
107 g/mol, d = 1.19 g/mL), toluene, ethanol, ethyl ether, dimethyl sulfoxide (DMSO), acetonitrile (ACN)
108 LC-MS grade, methanol (MeOH) LC-MS grade, sodium chloride (NaCl), anhydrous magnesium
109 sulphate (MgSO₄), sodium citrate dibasic sesquihydrate, sodium citrate tribasic dehydrate and PSA
110 sorbent were acquired from Scharlab (Barcelona, Spain). Ammonium acetate LC-MS grade and formic
111 acid LC-MS grade were purchased from Fluka (Busch, Switzerland). Ammonium fluoride was from
112 Panreac Química (Castellar del Vallès, Barcelona, Spain). A Millipore Milli-Q-System (Billerica, MA,
113 USA) was used for water (resistivity 18.2 MW cm).

114 Standards of PAs and PANOs with high purity grade ($\geq 90\%$) were supplied by PhytoLab GmbH &
115 Co. KG (Vestenbergsgreuth, Germany). Only retrorsine was acquired from Sigma-Aldrich (St. Louis,
116 MO, USA). Individual standard solutions (1000 $\mu\text{g/mL}$) were prepared according to the solubility of
117 each compound. Thus, intermedine, lycopsamine, retrorsine, seneciphylline, senecionine, heliotrine,
118 heliotrine *N*-oxide, europine and europine *N*-oxide were prepared in ACN/DMSO (4/1, v/v), whereas
119 senkirkin, senecivernine, senecivernine *N*-oxide, echimidine, echimidine *N*-oxide, lasiocarpine,
120 lasiocarpine *N*-oxide, intermedine *N*-oxide, lycopsamine *N*-oxide, retrorsine *N*-oxide, seneciphylline *N*-
121 oxide and senecionine *N*-oxide were prepared in MeOH. From the individual solutions, a standard
122 solution containing all the 21 analytes at 1 $\mu\text{g/mL}$ (each of them) was prepared in water. This
123 multicomponent solution was used to achieve working standard solutions at different concentration
124 levels by appropriate dilution with water to carry out the analytical performance of the method. All the
125 standard solutions were stored at -20 °C.

126 2.2. Synthesis and characterization of LP-MS and LP-MS-NH₂ mesostructured silicas

127 The large-pore mesostructured silica was synthesized as follows: Pluronic 123 (12 g) was dissolved
128 in 0.1% HCl (420 mL). The solution was stirred until homogenization at 30 °C. Then, ammonium
129 fluoride (0.14 g) was added and stirred for 10 min. Subsequently, TEOS and decane (25.84 g and 75

130 mL, respectively) were added, drop by drop, to the mixture, which was stirred for 20 h at 30 °C. After
131 this reaction time, the mixture was transferred with 25 mL of Milli-Q water to an autoclave reactor and
132 heated at 100 °C for 48 h. The solid product was recovered by filtration and washed with Milli-Q water.
133 Finally, it was calcinated at 540 °C for 8 h to remove the residual surfactant. The resultant material was
134 denoted as LP-MS. Subsequently, 3 g of LP-MS were weighted and mixed with toluene (40 mL). 3 mL
135 of (3-aminopropyl)triethoxysilane were added and the mixture was stirred at 80 °C for 24 h in a silicone
136 bath and under nitrogen atmosphere. The material was recovered by filtration and was washed
137 successively with toluene, ethanol and of ethyl ether (40 mL of each one). Finally, the material was dried
138 at 50-60 °C overnight and denoted as LP-MS-NH₂.

139 Both materials were characterized. The data and results related to the characterization assays are
140 included in the Supplementary Material (see SM1).

141 *2.3.Samples*

142 Dried aromatic herbs samples, including rosemary, basil, thyme, and herbs de Provence, with
143 different types of farming (conventional or organic) and from different geographical origins were
144 acquired at local supermarkets in Madrid (Spain). Sampling was performed according to the European
145 Commission Regulation No. 401/2006 concerning sampling and analysis of mycotoxins in foodstuff. In
146 this sense, 5 sub-samples were acquired for each lot number. Additionally, a rosemary sample and a
147 basil sample were collected from plants grown “in-house” in individual pots, from Toledo (Spain) and
148 Madrid (Spain), respectively. These samples were collected from the plants and dried for their analysis.
149 Also, a thyme sample on the branch was collected from a wild crop field in Cádiz (Spain). Sample details
150 are shown in Table S1 of the Supplementary Material. Samples were denoted by indicating in the first
151 letter the type of aromatic herb (*R* for rosemary, *B* for basil, *T* for thyme, and *H* for herbs de Provence)
152 followed by a dash with their type of farming (*C* for conventional, *O* for organic and *W* for wild farming,
153 whereas *I* was for samples collected from plants grown “in-house”). All samples were separately milled

154 to a fine powder with a grinder (A11 basic analytical mill, IKA® - Werke GmbH & Co. KG, Staufen,
155 Germany) for their homogenization and stored until their analysis. Each sample was analyzed in
156 triplicate.

157 *2.4. Modified μ -QuEChERS procedure*

158 The miniaturization of the QuEChERS procedure was based on a previous work of our research
159 group carried out for oregano samples (Izcara et al., 2020), with modifications: 1 mL of water was added
160 to 0.2 g of dry sample (previously weighted) for hydration of the sample matrix. This mixture was
161 vortexed for 1 min and then was magnetically stirred for 30 min. Subsequently, 1 mL of ACN was added
162 to the mixture, vortexed for 1 min and magnetically stirred for 30 min. Then, 0.65 g of the partitioning
163 salts mixture (MgSO₄, NaCl, sodium citrate tribasic dehydrate and sodium citrate dibasic sesquihydrate
164 in proportion 4:1:1:0.5) were added and vortexed for 1 min, followed by ultrasound agitation (5 min)
165 and centrifugation (10 min at 6000 rpm). The upper part of the supernatant corresponding to the ACN
166 fraction was separated and collected, while the rest of the sample extract was re-extracted again with 0.5
167 mL of ACN, vortexed for 1 min, ultrasound assisted (5 min) and centrifuged (10 min at 6000 rpm). The
168 aliquot from the upper part of the supernatant corresponding to the ACN fraction was collected with the
169 previous one and transferred to an Eppendorf containing 150 mg of MgSO₄ and 25 mg of the clean-up
170 sorbent (LP-MS, LP-MS-NH₂ or PSA). This mixture was vortexed for 1 min and centrifuged (5 min at
171 10,000 rpm). The supernatant was collected in a chromatographic vial, and the residue was re-extracted
172 again with 250 μ L of ACN, vortexed for 1 min and centrifuged (5 min at 10,000 rpm). The supernatant
173 was collected with the previous one in the vial and filtered through a 0.45 μ m PTFE filter membrane for
174 its subsequent injection in the chromatographic system.

175 *2.5. UHPLC-MS/MS analysis*

176 An UHPLC system (Dionex UltiMate 3000, Thermo Scientific, Waltham, MA, USA) coupled to an
177 ion-trap tandem mass spectrometer detector (ESI-ITMS amaZon SL, Bruker) was used for analysis.

178 Parameters for mass spectrometry acquisition were set as follows: electrospray ionization interface (ESI)
179 in positive ion mode, capillary voltage -4500 V, end plate offset -500 V, nebulizer gas 20 psi, dry gas
180 10 L/min and dry temperature at 200 °C. Multiple reaction monitoring (MRM) scan mode was used for
181 all analytes. The ESI source parameters for each analyte were determined by direct infusion of individual
182 standard solutions (5 $\mu\text{g/mL}$) at a flow rate of 4 $\mu\text{L/min}$. By individually infusing the analytes, it was
183 possible to identify the precursor ion of each analyte ($[\text{M}+\text{H}]^+$) in positive ion mode. Then, this precursor
184 ion was isolated and fragmented to obtain the corresponding product ions of each analyte. In this sense,
185 MS^2 was performed, and through the software of the mass spectrometer the extracted ion scan
186 chromatograms were obtained with the mass spectrum of each analyte. The chromatographic separation
187 of the 21 PAs/PANOs was carried out according to our previous work (Izcara et al., 2020), using a Luna
188 Omega Polar C18 column (100 mm x 2.1 mm, 1.6 μm particle size, Phenomenex, Torrance, CA, USA)
189 at 25 °C and a gradient elution. The mobile phase included water containing 0.2% formic acid and 5 mM
190 ammonium acetate (solvent A) and MeOH containing 10 mM ammonium acetate (solvent B). The
191 gradient conditions were: 5% B ($0-0.5$ min), $5-50\%$ B ($0.5-7$ min), 50% B ($7-7.5$ min), $50-100\%$ B
192 ($7.5-11$ min), 100% B ($11-12$ min), $100-5\%$ B ($12-14$ min) and re-equilibration of 1 min to initial
193 conditions, yielding a total analysis time of 15 min. The flow rate was 0.250 mL/min, and the injection
194 volume 2 μL . Retention time and mass spectrum parameters are presented in Table 1. For the
195 identification and confirmation of the compounds, the most intense product ion obtained for each analyte
196 in its mass spectrum (MS^2) was used for quantification, while the other product ions obtained were
197 monitored for confirmatory purposes.

198 *2.6. Assessment of analytical parameters*

199 Several analytical parameters were determined for method evaluation and method validation. These
200 parameters were assessed following the criteria described in the SANTE/12682/2019 document, in
201 regulation EC No 401/2006, and in the Q2(R1) ICH guidelines (International Council for Harmonisation
202 2005), since there is currently no official regulation for the validation of analytical methods to monitor

203 the presence of PAS/PANOs in food or feed. Accordingly, linearity was assessed with matrix-matched
204 calibration curves prepared in three consecutive days. These curves were prepared for each matrix at six
205 known concentration levels within the linear range evaluated. For this purpose, the sample extracts
206 obtained after the μ -QuEChERS procedure were spiked with an aliquot of a standard solution containing
207 the target analytes according to the desired concentration level of the calibration curve. Likewise, an
208 unspiked sample extract (denoted blank sample) also subjected to the μ -QuEChERS procedure was
209 analyzed in case some analytes were found in the sample in a natural way, so their signal could be
210 subtracted. The criteria for good linearity involve coefficient of determination (R^2) values closed to 1
211 and values $\leq \pm 20\%$ for the deviation of the back-calculated concentrations of the calibration standards
212 from the true concentrations (SANTE/12682/2019; EC No 401/2006). Matrix effects were determined
213 by comparing the slopes of the calibration equations obtained from both matrix-matched and solvent-
214 based calibration curves (both expressed in the same units $\mu\text{g/L}$), calculating the ratio slope matrix-
215 matched/slope solvent-based*100 for each of the 21 analytes. A ratio lower than 100% suggests signal
216 suppression, whereas a ratio greater than 100% indicates signal increase. When the value is in the range
217 of 80-120%, the matrix effects can be ignored. However, when the signal suppression or enhancement
218 is greater than this margin of 20%, matrix effects must be considered in calibration (European
219 Commission SANTE/12682/2019). Nonetheless, values within a margin $\pm 40\%$ could be determined as
220 soft matrix effects, but they need to be considered in calibration.

221 The selectivity of the method was determined by comparing the spectra of the different analytes
222 obtained from standard solutions with the spectra obtained in the samples. It was considered satisfactory
223 when the variation in the spectra was less than $\pm 30\%$ and the retention time of the analytes was within
224 the interval $\pm 2.5\%$ (European Commission SANTE/12682/2019). The sensitivity of the method for
225 each matrix was determined through the method detection limits (MDLs) and method quantification
226 limits (MQLs) of the analytes. These limits were estimated based on the standard deviation of the

227 response and the slope obtained in the matrix-matched calibration curves for the lowest concentration
228 level (Q2(R1) International Council for Harmonisation, 2005):

229 $MDL = 3.3 \times \text{standard deviation of the response at the lowest concentration} / \text{slope of the calibration}$
230 curve

231 $MQL = 10 \times \text{standard deviation of the response at the lowest concentration} / \text{slope of the calibration curve}$

232 The recovery assays were assessed by comparing the areas obtained for samples spiked with a known
233 concentration of analytes and subjected to the μ -QuEChERS procedure with those areas obtained for
234 simulated samples (samples spiked at the same concentration but at the end of the μ -QuEChERS
235 procedure prior to their chromatographic analysis). For method evaluation, the recovery assays were
236 performed spiking the aromatic herb samples at a concentration of 100 $\mu\text{g/Kg}$. A maximum
237 concentration of 400 $\mu\text{g/Kg}$ of total PAs and PANOs has been established for dried herbs (excluding
238 borage, lovage, marjoram and oregano, for which a maximum concentration of 1000 $\mu\text{g/Kg}$ is set)
239 (COMMISSION REGULATION (EU) 2020/2040). Therefore, for method validation, the accuracy was
240 evaluated in terms of recovery for the aromatic herb matrices at three concentration levels: low (10
241 $\mu\text{g/Kg}$), medium (100 $\mu\text{g/Kg}$) and high (800 $\mu\text{g/Kg}$), so that this value can be covered in a wide range.
242 These results were expressed as the mean recovery obtained from six samples ($n = 6$) spiked with the
243 analytes at the corresponding concentration (low, medium or high) and subjecting them to the proposed
244 extraction procedure. According to the validation guidelines, the recovery values should be between 70
245 and 120% (SANTE/12682/2019; EC No 401/2006). On the other hand, the method precision was
246 evaluated in terms of repeatability and reproducibility, using the same validation levels (low, medium
247 and high) than for the accuracy. For repeatability (expressed as RSD%), a sample spiked with the
248 analytes at the corresponding validation level was consecutively injected six times ($n = 6$) on the same
249 day. The reproducibility (also expressed as RSD%) was calculated by the analysis of three replicates of
250 a sample (spiked with the analytes at the corresponding validation level), which were injected in

251 triplicate throughout three different days ($n = 9$). According to the validation guidelines, the RSD values
252 for these precision parameters should be $\leq 20\%$ (SANTE/12682/2019; EC No 401/2006).

253 The validation of the method was carried out for each matrix using a representative sample of each
254 of the aromatic herbs. As no blank samples or certified materials were available, the validation was
255 carried out with samples R-C-1, B-C-1 and T-C-1 for rosemary, basil and thyme matrices, respectively.
256 As the herbs de Provence are a mixture of different aromatic herbs (Table S1), this type of matrix was
257 not validated. Nevertheless, some analytical parameters of this matrix were assessed with sample H-O-
258 2, such as linearity, MDLs, MQLs, and the accuracy at one level (Table S2).

259 *2.7. Statistical Analysis*

260 Each aromatic herb sample was analyzed in triplicate. The statistical analysis of the samples was
261 performed with SPSS 19.0 software, using one-way analysis of variance (ANOVA) and Duncan multiple
262 range test (significant differences at $p \leq 0.05$).

263 **3. Results and discussion**

264

265 *3.1. Evaluation of clean-up sorbents and optimization of the μ -QuEChERS procedure*

266 The mesostructured silicas synthesized (LP-MS and LP-MS-NH₂) and a commercial PSA were
267 evaluated as clean-up sorbents under same conditions. For this purpose, these materials were initially
268 tested in the clean-up step of the μ -QuEChERS procedure described in our previous work (Izcara et al.,
269 2020). The miniaturization of this method provides a cost-effective and environmentally friendly
270 microextraction that enables to improve the original QuEChERS strategy, as it successfully reduces the
271 amounts of sample, solvents, clean-up sorbents and partitioning salts employed, as previously described
272 in Section 2.4. Accordingly, after analytes were extracted with 1 mL of ACN followed by partitioning
273 of the analytes in the presence of a salt mixture (MgSO₄, NaCl, sodium citrate tribasic dehydrate and
274 sodium citrate dibasic sesquihydrate in proportion 4:1:1:0.5), the ACN phase was further cleaned up

275 with anhydrous MgSO₄ and 25 mg of clean-up sorbent (PSA, LP-MS or LP-MS-NH₂). The clean-up
276 efficiency of the materials was assayed through the determination of matrix effects and the method
277 recoveries as explained in Section 2.6. These preliminary studies were carried out with rosemary and
278 basil (R-C-1 and B-C-1, respectively). As shown in Figure 2a and Table S3 most of the compounds
279 presented significant matrix effects in both matrices with the three materials tested, as they showed
280 values lower than 80%. In general, matrix effects were stronger with PSA than with LP-MS and LP-
281 MS-NH₂ in both samples (Figure 2a, Table S3), suggesting that the clean-up efficiency of
282 mesostructured silicas was higher than the one achieved with PSA. Therefore, it was confirmed that
283 mesostructured silicas can be useful clean-up sorbents. Regarding both silicas, LP-MS-NH₂ provided
284 less matrix effects than LP-MS, as more analytes had values in the range 80-120% (Figure 2a, Table
285 S3). Therefore, LP-MS-NH₂ seemed to be the most suitable clean-up sorbent, as fewer analytes were
286 affected by the matrix interferences when using this material (Figure 2a, Table S3). Nevertheless,
287 regarding the recovery values, although they followed a similar trend among the three materials in the
288 rosemary matrix, for some analytes the recoveries were low (<60%), such as for: intermedine, retrorsine,
289 europine, lycopsamine, senkirkin, intermedine *N*-oxide, seneciophylline, echimidine *N*-oxide, europine
290 *N*-oxide and lycopsamine *N*-oxide (Figure S1). The recoveries in basil were also similar among the three
291 sorbents, but some of them were also below the 60% (intermedine, europine, lycopsamine and their
292 corresponding *N*-oxides) (Figure S2). These low recovery values observed for some of the analytes were
293 generally obtained with all three sorbents in both matrices (Figures S1 and S2). It was observed that the
294 most polar analytes (those that eluted in the chromatogram first) were the ones with the lowest recovery
295 values, but at the same time were the ones with the less matrix effects. This may suggest that the sorbent
296 materials employed have more affinity to retain interferences of polar type because of the polar
297 properties of the chemical structure of the silicas. In fact, PSA is indicated for removing polar organic
298 acids, polar pigments and sugars. Therefore, the most polar PAs can also be retained in the material, and
299 for this reason, probably lower recovery values were achieved for them. Overall, despite the low

300 recovery values achieved for some analytes, as LP-MS-NH₂ seemed to provide less matrix effects in
301 both samples (Figure 2a, Table S3), it was selected as the most suitable clean-up sorbent (Figure 2a,
302 Table S3). Nonetheless, with the aim of improving the recovery values, some modifications in the μ -
303 QuEChERS procedure were performed for its optimization. These modifications included a second
304 extraction cycle with 0.5 mL of ACN before the clean-up procedure and an elution step with 250 μ L of
305 ACN after the clean-up step, as explained in section 2.4. As shown in Figure 2b and c, with these
306 modifications the recoveries in the rosemary and basil samples were significantly improved in almost
307 all the analytes. The overall recoveries for rosemary with the new extraction conditions ranged from 67
308 to 107% (Figure 2b), whereas in the basil matrix they ranged from 78 to 105% (Figure 2c).

309 Finally, under these final extraction conditions, the method recoveries were also assayed for a thyme
310 sample (T-C-1) and an herb de Provence sample (H-O-2) (Figure S3). As it can be observed, good
311 recoveries were achieved for all the analytes in the thyme sample, ranging from 75 to 106% (Figure
312 S3a). In contrast, the recoveries in the herbs de Provence matrix range from 79 to 103% (Figure S3b),
313 Therefore, in general, good recovery values were achieved for all the PAs/PANOs analyzed in the
314 different herb matrices. Thus, it was confirmed that LP-MS-NH₂ could be effectively used as clean-up
315 sorbent in the different aromatic herb samples.

316 *3.2. Method validation*

317 Once LP-MS-NH₂ was selected as the most efficient clean-up sorbent, the method was validated in
318 terms of linearity, selectivity, MDLs, MQLs, accuracy and precision. Good analytical performance of
319 the method was achieved for the three aromatic herbs. As it can be observed in Tables 2-4, all compounds
320 showed good linear regression, with coefficient of determination (R^2) values > 0.998 . Moreover, values
321 (%) of the deviation of the back-calculated concentrations ranged from -11 to +19% for rosemary and
322 basil, and from -20 to +19% for thyme. Therefore, this parameter was successfully achieved in the three
323 matrices, as the values were in all cases $\leq \pm 20\%$ (SANTE/12682/2019). In addition, the deviation (as

324 RSD%) of the slopes of the matrix-matched calibration curves prepared in three consecutive days ranged
325 from 0.5 to 8%. Good selectivity of the method was achieved, as the deviation of the ion ratios obtained
326 in the different samples did not deviate more than $\pm 30\%$ in comparison to the mass spectra obtained
327 with standard solutions. Moreover, the retention time of all the analytes was within the interval $\pm 2.5\%$.
328 MDLs of the analytes were in the range 0.7-3.0, 0.7-3.0 and 0.4-2.9 $\mu\text{g/Kg}$, and MQLs 2.5-9.9, 2.2-10.0
329 and 1.2-9.7 $\mu\text{g/Kg}$ for rosemary, basil and thyme, respectively (Table 2-4). On the other hand, the overall
330 average recoveries obtained for the three validation levels were in the range 79-103%, 88-103% and 73-
331 105% for rosemary, basil and thyme respectively (Tables 2-4). Therefore, this validation parameter was
332 successfully accomplished as all the values were within the range 70-120% (SANTE/12682/2019; EC
333 No 401/2006). Likewise, satisfactory precision values were obtained at the three validation levels in the
334 three matrices, as all of them were $\leq 20\%$ (Tables 2-4). Therefore, as the validation guidelines were fully
335 accomplished, the analytical performance of the $\mu\text{-QuEChERS}$ procedure proposed was successfully
336 demonstrated. Thus, this procedure can be reliably applied to the analysis of PAs and PANOs in aromatic
337 herb samples.

338 *3.3. Analysis of samples*

339 The feasibility of the method was demonstrated by its application to the analysis of 17 samples,
340 including 4 rosemary samples, 5 basil samples, 4 thyme samples and 4 herbs de Provence samples
341 (Figure 3). The quantification was performed with the matrix-matched calibration curves calculated for
342 each type of aromatic herb matrix. Contents below the MDL were considered as 0.0 $\mu\text{g/Kg}$ (not
343 detected), whereas contents between the MDL and the MQL were included as $<\text{MQL}$ (Table S4). As it
344 can be observed in Figure 3, all the samples analyzed were contaminated with PAs and PANOs, but all
345 the 21 target analytes were not always found in all the samples. According to COMMISSION
346 REGULATION (EU) 2020/2040, the maximum amount of PAs/PANOs allowed in dried herbs (except
347 for borage, lovage, marjoram and oregano) is 400 $\mu\text{g/Kg}$. Accordingly, all the samples analyzed were
348 below this limit, except two thyme samples: T-C-1 (447 $\mu\text{g/Kg}$) and T-O-1 (553 $\mu\text{g/Kg}$) (Figure 3a).

349 Conversely, the smallest average content (49 $\mu\text{g}/\text{Kg}$) was found in a rosemary sample (R-C-2) (Figure
350 3a). Based on the structural and botanical origin, PAs/PANOs can be classified in four different families
351 (heliotrine-type, senecionine-type, lycopsamine-type and monocrotaline-type) (Picron et al., 2018a).
352 Accordingly, heliotrine-type PAs (particularly, lasiocarpine, lasiocarpine *N*-oxide and europine *N*-oxide)
353 were the ones which significantly contributed to the contamination of the aromatic herb samples
354 analyzed, as they were often found in the samples at a relatively higher concentration value than the
355 other PAs/PANOs, followed by the senecionine-type PAs (mainly, senecivernine *N*-oxide and
356 senecionine *N*-oxide) (Figure 3b). This contamination profile matches with the one described by other
357 authors in previous works (Picron et al., 2018a; Kaltner et al., 2020). The occurrence of heliotrine-type
358 compounds is usually related to co-harvesting or adulteration with *Heliotropium spp.* and *Borago spp.*,
359 whereas the contamination with senecionine-type PAs is often associated to species of the Asteraceae
360 family, mainly *Senecio vulgaris* (Picron et al., 2018a; Kapp, Hägele, and Plate, 2019; Kaltner et al.,
361 2020). Regarding lycopsamine-type PAs, only the occurrence of echimidine was relevant in some of the
362 samples analyzed (R-C-3, T-C-1, T-W-1, H-C-1 and H-C-2) (Figure 3b), which may indicate
363 contamination with plants belonging to the Boraginaceae family, such as *Borago spp* (Kaltner et al.,
364 2020; Mädge et al., 2020). Thyme and basil samples were the most contaminated samples, with an
365 average content of PAs/PANOs of 394.25 and 293.40 $\mu\text{g}/\text{Kg}$, respectively. In contrast, rosemary samples
366 were the least contaminated, with an average content of PAs/PANOs of 148.25 $\mu\text{g}/\text{Kg}$ (Figure 3a). The
367 herbs de Provence are a mixture of different aromatic herbs (Table S1), so their contamination can be
368 due to more than one aromatic herb. In this sense, this mixture of herbs also contains oregano, which is
369 one of the culinary herbs for which most of the food alerts related to concerning high values of
370 PAs/PANOs have been notified (Izcara et al., 2020). Accordingly, several works in the literature have
371 reported high levels of PAs and PANOs in oregano samples (Kapp, Hägele, and Plate, 2019; Kaltner et
372 al., 2020; Izcara et al., 2020). However, despite their content in oregano and other herbs (Table S1), the

373 herbs de Provence samples analyzed in this work showed an average value of PAs/PANOs of 203.50
374 $\mu\text{g/Kg}$, which is lower than the ones obtained for other herbs such as thyme and basil (Figure 3a).

375 The samples grown “in-house” in a private garden were expected not be contaminated. However,
376 these samples were also positive, although, in general, they were less contaminated than the samples
377 acquired from the supermarket (Figure 3a). The contamination pattern among these sample was very
378 similar, with senecivernine *N*-oxide, lasiocarpine and europine *N*-oxide as the main compounds that
379 contributed to their contamination (Figure 3b). The occurrence of these alkaloids in the “in-house”
380 samples reinforces the horizontal natural transfer of PAs/PANOs through the soil among living plants
381 growing nearby or from dead plant materials (Selmar et al., 2015; Nowak et al., 2016; Selmar et al.,
382 2019), since the soil and compost employed in the pots of these plants had previously been used to grow
383 other types of plants, which could be PAs-producing plants or be contaminated with weeds containing
384 PAs /PANOs. Moreover, the thyme sample collected from a wild field showed significantly higher
385 PAs/PANOs contamination values than the samples “in-house” (Figure 3a), probably because it was
386 more exposed to fields of PAs-producing plants growing nearby, what reassert the horizontal natural
387 transfer as contamination path of PAs/PANOs. In fact, in this wild sample the occurrence of some
388 lycopsamine-type alkaloids, such as lycopsamine *N*-oxide and intermedine *N*-oxide, stood out compared
389 to the other samples analyzed, in which in most of them these compounds were not even present (Figure
390 3b). This highlights the wide variety of unexpected botanical species that may contaminate these herbs.

391 Regarding the type of farming, it was not possible to draw significant conclusions. Among the basil
392 samples, no significant differences were observed between the samples produced by conventional and
393 organic farming, as the total amount of PAs/PANOs was very similar among them (Figure 3a). It was
394 only noticed that seneciophylline *N*-oxide was only found in the conventional farming samples, whereas
395 europine was only in the organic farming samples (Figure 3b). In the case of the herbs de Provence
396 samples, the ones obtained by organic farming were less contaminated than the ones with conventional
397 farming (Figure 3a). However, in the thyme samples, the sample most contaminated was one produced

398 by organic farming (T-O-2), which in fact was the sample that presented the highest contamination value
399 (553 $\mu\text{g}/\text{Kg}$) of all the aromatic herbs analyzed (Figure 3a). Moreover, in general, all the aromatic herbs
400 presented the same contamination profile regardless of their type of farming (Figure 3b).

401 **4. Conclusions**

402 The original QuEChERS strategy was successfully miniaturized by reducing the amounts of sample
403 (0.2 g), solvents (2 mL), clean-up sorbents (25 mg sorbent + 150 mg MgSO_4) and partitioning salts (0.65
404 g) employed, leading to an improved cost-effective and environmentally friendly microextraction
405 method, which meets the Green Analytical Chemistry principles. Moreover, it was confirmed that
406 mesostructured silicas could be considered as promising and alternative clean-up sorbents in sample
407 preparation. The feasibility of the method proposed with LP-MS- NH_2 was determined by its validation
408 and its application to the analysis of 17 different aromatic herbs. All the samples analyzed were
409 contaminated with PAs and PANOs, but only in two thyme samples the sum of the total PAs/PANOs
410 exceeded 400 $\mu\text{g}/\text{Kg}$, which is the maximum limit regulated for these compounds in aromatic herbs. In
411 general, all the aromatic herbs presented the same contamination profile regardless of their type of
412 farming. Heliotrine-type PAs were the ones which significantly contributed to the contamination of the
413 aromatic herb samples analyzed, followed by the senecionine-type PAs, whereas the occurrence of
414 lycopsamine-type PAs was less significant. In this sense, lasiocarpine, europine *N*-oxide and
415 senecivernine *N*-oxide were the PAs that significantly contributed to the contamination of the samples
416 analyzed. In addition, the horizontal natural transfer of PAs/PANOs through the soil among living plants
417 growing nearby or from dead plant materials was reinforced as possible contamination path through the
418 analysis of samples cultivated “in-house” and collected from wild fields. Overall, this work confirmed
419 the concerning occurrence of these contaminants in aromatic herbs, highlighting the need to develop
420 analytical strategies that enable to monitor and regulate the presence of these contaminants in food items
421 to ensure the safety of consumers.

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424 S.I., N.C., S.M.-Z. and D.P-Q.; writing—original draft preparation, N.C.; writing—review and editing,
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550

551 **Figure Captions**

552 **Fig. 1** Pyrrolizidine alkaloids included in this work and classified into different families based on their
553 structural similarities and botanical origin.

554 **Fig. 2 (a)** 2D plot of the matrix effect of the analytes/retention time obtained in rosemary and basil
555 samples using PSA, LP-MS and LP-MS-NH₂ as clean-up sorbents. Recovery percentages obtained from
556 the analysis of **(b)** three spiked replicates of a rosemary sample (100 µg/Kg of each analyte) and **(c)** three
557 spiked replicates of a basil sample (100 µg/Kg of each analyte), extracted by the modified µ-QuEChERS
558 procedure proposed before and after its optimization. Error bars represent the standard deviation of
559 sample replicates (n = 3).

560 **Fig. 3 (a)** Heat map plot of the individual content of PAs and PANOs and **(b)** total content of
561 PAs/PANOs (µg/Kg) found in the different aromatic herb samples analyzed by the modified µ-
562 QuEChERS method proposed. In the sample identification code, the first letter indicates: *R* for rosemary,
563 *B* for basil, *T* for thyme, and *H* for herbs de Provence; followed by their type of farming: *C* for
564 conventional, *O* for organic and *W* for wild farming, whereas *I* was for samples collected from plants
565 grown “in-house”.

Table 1. Retention time and mass spectrum parameters of the targeted analytes using the UHPLC-IT-MS/MS method developed in positive ESI ion mode.

Analyte	Retention Time (min)	Precursor Ion (<i>m/z</i>)	Fragmentation Amplitude	MS ² . Product Ions ^a (<i>m/z</i>)
Intermedine	5.6	299	0.70	138 , 120
Europine	5.7	329	0.80	253 , 138
Lycopsamine	5.8	299	0.70	138 , 120
Europine <i>N</i> -oxide	6.2	345	0.80	327 , 171.5
Intermedine <i>N</i> -oxide	6.4	315	0.80	225, 171.5
Lycopsamine <i>N</i> -oxide	6.5	315	0.80	171.5 , 138
Retrorsine	6.8	351	0.80	323 , 275
Retrorsine <i>N</i> -oxide	7.0	367	0.90	339 , 245
Seneciphylline	7.2	333	0.80	305 , 120
Heliotrine	7.2	313.5	0.70	138 , 120
Seneciphylline <i>N</i> -oxide	7.5	350	0.80	321 , 118
Heliotrine <i>N</i> -oxide	7.6	329	1.00	171 , 136
Senecivernine	7.9	335	0.80	307 , 120
Senecionine	7.9	335	0.80	307 , 120
Senecivernine <i>N</i> -oxide	8.1	351	0.80	323 , 219.5
Senecionine <i>N</i> -oxide	8.3	352	1.00	220, 118
Echimidine	8.7	398	0.60	220, 120
Echimidine <i>N</i> -oxide	8.7	413	0.70	395 , 351
Senkirkin	9.1	365	0.80	167.5 , 150
Lasiocarpine	9.8	411	0.70	335 , 219.5
Lasiocarpine <i>N</i> -oxide	10.4	428	0.80	409 , 352

^a Predominant product ion. Ions in bold were the ones used for quantification. Isolation width (*m/z*) is 4. Retention time with the optimized gradient elution: 5% B (0–0.5 min), 5–50% B (0.5–7 min), 50% B (7–7.5 min), 50–100% B (7.5–11 min), 100% B (11–12 min), 100–5% B (12–14 min). Water containing 0.2% formic acid and 5 mM ammonium acetate as mobile phase A and methanol containing 10 mM ammonium acetate as mobile phase B. The flow rate was 0.25 mL/min.

Table 2. Validation parameters of the modified μ -QuEChERS method proposed for the determination of the target PAs/PANOs in rosemary samples.

Analytes	Linear Range ($\mu\text{g/Kg}$)	Matrix-Matched Calibration R^2	Accuracy		Precision		MDL ($\mu\text{g/Kg}$)	MQL ($\mu\text{g/Kg}$)
			Recovery (% \pm sd)	Mean Recovery (% \pm sd)	Repeatability (RSD%)	Reproducibility (RSD%)		
Intermedine	10.0–500.0	$y = 10118x - 56959$ 0.999	82 ± 7^a	90 ± 8	14^a	17^a	2.7	9.1
			98 ± 2^b		9^b	11^b		
			89 ± 3^c		5^c	11^c		
Europine	10.0–500.0	$y = 4796x - 51720$ 0.998	101 ± 7^a	80 ± 18	13^a	16^a	1.2	4.0
			71 ± 2^b		5^b	16^b		
			69 ± 4^c		8^c	9^c		
Lycopsamine	10.0–500.0	$y = 5661x - 39739$ 0.999	104 ± 4^a	79 ± 21	10^a	10^a	2.2	7.2
			67 ± 0^b		8^b	12^b		
			67 ± 2^c		8^c	12^c		
Europine <i>N</i> -oxide	10.0–500.0	$y = 13771x - 233753$ 0.999	100 ± 2^a	97 ± 3	9^a	20^a	2.7	9.1
			94 ± 1^b		6^b	12^b		
			96 ± 5^c		7^c	9^c		
Intermedine <i>N</i> -oxide	10.0–500.0	$y = 9911x - 131545$ 0.998	101 ± 3^a	93 ± 11	11^a	16^a	2.5	8.3
			98 ± 3^b		5^b	7^b		
			80 ± 2^c		5^c	8^c		
Lycopsamine <i>N</i> -oxide	10.0–500.0	$y = 5899x - 70610$ 0.998	99 ± 3^a	91 ± 12	9^a	14^a	1.6	5.3
			96 ± 0^b		11^b	11^b		
			77 ± 1^c		14^c	16^c		
Retrorsine	10.0–500.0	$y = 86x + 11503$ 0.999	103 ± 1^a	101 ± 7	14^a	19^a	2.9	9.6
			107 ± 2^b		14^b	16^b		
			94 ± 4^c		7^c	19^c		
Retrorsine <i>N</i> -oxide	10.0–500.0	$y = 955x - 11899$ 0.999	101 ± 4^a	103 ± 3	8^a	9^a	2.7	9.2
			106 ± 1^b		18^b	19^b		
			102 ± 6^c		8^c	17^c		
Seneciphylline	10.0–500.0	$y = 2129x - 5999$ 0.999	98 ± 4^a	86 ± 12	13^a	16^a	2.8	9.4
			87 ± 9^b		13^b	15^b		
			74 ± 0^c		11^c	17^c		
Heliotrine	10.0–500.0	$y = 22260x + 82018$ 0.999	93 ± 2^a	95 ± 2	17^a	17^a	0.7	2.5
			96 ± 0^b		16^b	16^b		
			95 ± 4^c		10^c	16^c		
Seneciphylline <i>N</i> -oxide	10.0–500.0	$y = 1314x + 17526$ 0.999	103 ± 5^a	99 ± 3	6^a	15^a	1.3	4.3
			98 ± 2^b		3^b	16^b		
			97 ± 2^c		12^c	19^c		
Heliotrine <i>N</i> -oxide	10.0–500.0	$y = 135x + 1294$ 0.999	102 ± 4^a	93 ± 8	5^a	16^a	2.1	6.9
			86 ± 3^b		7^b	15^b		
			91 ± 2^c		14^c	17^c		

Senecivernine	10.0–500.0	$y = 2530x - 49696$ 0.999	101 ± 5^a 100 ± 2^b 78 ± 2^c	93 ± 13	12^a 12^b 11^c	19^a 15^b 15^c	1.2	4.2
Senecionine	10.0–500.0	$y = 2813x - 26048$ 0.999	103 ± 3^a 100 ± 4^b 84 ± 0^c	96 ± 10	19^a 12^b 10^c	20^a 18^b 15^c	2.4	8.2
Senecivernine <i>N</i> -oxide	10.0–500.0	$y = 1825x + 2034$ 0.999	93 ± 1^a 93 ± 0^b 87 ± 1^c	91 ± 3	5^a 10^b 12^c	13^a 16^b 19^c	1.7	5.7
Senecionine <i>N</i> -oxide	10.0–500.0	$y = 4247x - 40862$ 0.999	102 ± 4^a 100 ± 0^b 73 ± 3^c	92 ± 16	12^a 10^b 12^c	19^a 15^b 12^c	2.5	8.5
Echimidine	10.0–500.0	$y = 85x + 2046$ 0.999	96 ± 5^a 92 ± 4^b 99 ± 6^c	96 ± 4	11^a 12^b 9^c	18^a 20^b 10^c	2.9	9.6
Echimidine <i>N</i> -oxide	10.0–500.0	$y = 2012x + 903405$ 0.998	98 ± 1^a 102 ± 2^b 101 ± 2^c	100 ± 2	16^a 14^b 6^c	19^a 14^b 19^c	3.0	9.9
Senkirkin	10.0–500.0	$y = 1293x + 519$ 0.999	96 ± 2^a 97 ± 3^b 71 ± 2^c	88 ± 15	19^a 8^b 12^c	20^a 18^b 13^c	2.6	8.6
Lasiocarpine	10.0–500.0	$y = 1396x - 40299$ 0.999	91 ± 2^a 99 ± 2^b 99 ± 1^c	96 ± 5	18^a 15^b 9^c	19^a 16^b 11^c	1.5	5.0
Lasiocarpine <i>N</i> -oxide	10.0–250.0	$y = 4399x - 44162$ 0.999	98 ± 2^a 99 ± 0^b 80 ± 3^c	92 ± 11	11^a 5^b 11^c	19^a 8^b 14^c	2.8	9.3

Recovery: mean recovery obtained from six samples ($n = 6$) spiked with the analytes at a known concentration level, and subjected to the proposed extraction procedure; Repeatability: six consecutive injections ($n = 6$) on the same day of a sample spiked with the analytes at a known concentration level; Reproducibility: three replicates of a sample injected in triplicate throughout three different days ($n = 9$) and spiked with the analytes at a known concentration level; MDL: method detection limit; MQL: method quantification limit; ^a Low spiked level (10 µg/Kg); ^b Medium spiked level (100 µg/Kg); ^c High spiked level (800 µg/Kg).

Table 3. Validation parameters of the modified μ -QuEChERS method proposed for the determination of the target PAs/PANOs in basil samples.

Analytes	Linear Range ($\mu\text{g/Kg}$)	Matrix-Matched Calibration R^2	Accuracy		Precision		MDL ($\mu\text{g/Kg}$)	MQL ($\mu\text{g/Kg}$)
			Recovery (% \pm sd)	Mean Recovery (% \pm sd)	Repeatability (RSD%)	Reproducibility (RSD%)		
Intermedine	10.0–500.0	$y = 10259x - 130642$ 0.999	101 \pm 2 ^a 80 \pm 2 ^b 96 \pm 5 ^c	92 \pm 11	18 ^a 3 ^b 8 ^c	18 ^a 15 ^b 18 ^c	3.0	9.8
Europine	10.0–500.0	$y = 6139x - 12675$ 0.999	99 \pm 4 ^a 92 \pm 7 ^b 103 \pm 1 ^c	98 \pm 6	14 ^a 11 ^b 7 ^c	15 ^a 17 ^b 15 ^c	2.4	7.9
Lycopsamine	10.0–500.0	$y = 6337x - 102712$ 0.999	97 \pm 2 ^a 78 \pm 4 ^b 89 \pm 1 ^c	88 \pm 10	16 ^a 13 ^b 6 ^c	18 ^a 17 ^b 14 ^c	3.0	10.0
Europine <i>N</i> -oxide	10.0–500.0	$y = 12199x - 47148$ 0.999	100 \pm 5 ^a 78 \pm 6 ^b 101 \pm 0 ^c	93 \pm 13	11 ^a 3 ^b 8 ^c	16 ^a 12 ^b 17 ^c	2.9	9.6
Intermedine <i>N</i> -oxide	10.0–500.0	$y = 12283x - 103692$ 0.999	99 \pm 1 ^a 85 \pm 4 ^b 82 \pm 7 ^c	89 \pm 9	10 ^a 5 ^b 8 ^c	15 ^a 10 ^b 9 ^c	1.5	5.1
Lycopsamine <i>N</i> -oxide	10.0–500.0	$y = 7959x - 93111$ 0.999	97 \pm 2 ^a 86 \pm 10 ^b 81 \pm 1 ^c	88 \pm 8	4 ^a 15 ^b 7 ^c	20 ^a 18 ^b 14 ^c	0.7	2.2
Retrorsine	10.0–500.0	$y = 95x + 1349$ 0.999	104 \pm 6 ^a 103 \pm 3 ^b 101 \pm 1 ^c	103 \pm 2	6 ^a 11 ^b 7 ^c	15 ^a 20 ^b 15 ^c	2.8	9.3
Retrorsine <i>N</i> -oxide	10.0–500.0	$y = 198x + 3626$ 0.999	99 \pm 6 ^a 100 \pm 5 ^b 100 \pm 5 ^c	100 \pm 1	17 ^a 18 ^b 15 ^c	19 ^a 19 ^b 18 ^c	2.8	9.5
Seneciphylline	10.0–500.0	$y = 2647x - 14743$ 0.999	98 \pm 1 ^a 96 \pm 2 ^b 97 \pm 7 ^c	97 \pm 1	15 ^a 10 ^b 8 ^c	15 ^a 11 ^b 17 ^c	2.5	8.3
Heliotrine	10.0–500.0	$y = 31227x - 85693$ 0.999	102 \pm 4 ^a 105 \pm 1 ^b 98 \pm 4 ^c	102 \pm 4	13 ^a 7 ^b 13 ^c	17 ^a 17 ^b 19 ^c	2.0	6.6
Seneciphylline <i>N</i> -oxide	10.0–500.0	$y = 2950x - 40067$ 0.999	95 \pm 4 ^a 91 \pm 4 ^b 78 \pm 6 ^c	88 \pm 9	18 ^a 9 ^b 13 ^c	18 ^a 16 ^b 19 ^c	1.1	3.8
Heliotrine <i>N</i> -oxide	10.0–500.0	$y = 486x + 11088$ 0.999	96 \pm 5 ^a 101 \pm 3 ^b 91 \pm 6 ^c	96 \pm 5	19 ^a 14 ^b 18 ^c	20 ^a 17 ^b 20 ^c	2.8	9.2

Senecivernine	10.0–500.0	$y = 2094x + 3564$ 0.999	102 ± 5^a 99 ± 3^b 96 ± 1^c	99 ± 3	13^a 11^b 6^c	15^a 19^b 11^c	2.5	8.3
Senecionine	10.0–500.0	$y = 2724x - 24748$ 0.999	101 ± 4^a 99 ± 6^b 100 ± 6^c	100 ± 1	11^a 10^b 12^c	15^a 12^b 16^c	3.0	10.0
Senecivernine <i>N</i> -oxide	10.0–500.0	$y = 1626x + 309$ 0.999	98 ± 4^a 95 ± 2^b 76 ± 1^c	90 ± 12	18^a 10^b 15^c	18^a 15^b 17^c	2.8	9.5
Senecionine <i>N</i> -oxide	10.0–500.0	$y = 12218x - 218973$ 0.999	101 ± 3^a 98 ± 3^b 95 ± 6^c	98 ± 3	7^a 7^b 10^c	17^a 14^b 12^c	2.9	9.7
Echimidine	10.0–500.0	$y = 126x + 6839$ 0.999	97 ± 10^a 99 ± 2^b 101 ± 2^c	99 ± 2	9^a 16^b 14^c	18^a 17^b 20^c	1.3	4.3
Echimidine <i>N</i> -oxide	10.0–500.0	$y = 888x - 5278$ 0.999	101 ± 3^a 103 ± 3^b 103 ± 2^c	102 ± 1	15^a 12^b 14^c	16^a 19^b 16^c	1.7	5.8
Senkirkin	10.0–500.0	$y = 4607x - 9869$ 0.999	96 ± 1^a 98 ± 2^b 103 ± 4^c	99 ± 4	19^a 4^b 5^c	20^a 7^b 8^c	3.0	10.0
Lasiocarpine	10.0–500.0	$y = 1978x - 21551$ 0.999	104 ± 5^a 97 ± 5^b 103 ± 3^c	101 ± 4	4^a 11^b 13^c	19^a 18^b 18^c	1.2	3.9
Lasiocarpine <i>N</i> -oxide	10.0–500.0	$y = 37278x - 101868$ 0.999	100 ± 2^a 93 ± 4^b 93 ± 4^c	95 ± 4	11^a 3^b 5^c	11^a 10^b 9^c	2.8	9.3

Recovery: mean recovery obtained from six samples ($n = 6$) spiked with the analytes at a known concentration level, and subjected to the proposed extraction procedure; Repeatability: six consecutive injections ($n = 6$) on the same day of a sample spiked with the analytes at a known concentration level; Reproducibility: three replicates of a sample injected in triplicate throughout three different days ($n = 9$) and spiked with the analytes at a known concentration level; MDL: method detection limit; MQL: method quantification limit; ^a Low spiked level (10 µg/K); ^b Medium spiked level (100 µg/Kg); ^c High spiked level (800 µg/Kg).

Table 4. Validation parameters of the modified μ -QuEChERS method proposed for the determination of the target PAs/PANOs in thyme samples.

Analytes	Linear Range ($\mu\text{g/Kg}$)	Matrix-Matched Calibration R^2	Accuracy		Precision		MDL ($\mu\text{g/Kg}$)	MQL ($\mu\text{g/Kg}$)
			Recovery (% \pm sd)	Mean Recovery (% \pm sd)	Repeatability (RSD%)	Reproducibility (RSD%)		
Intermedine	10.0–500.0	$y = 7744x - 117703$ 0.999	80 ± 8^a	88 ± 8	8^a	17^a	2.9	9.7
			95 ± 1^b		11^b	16^b		
			90 ± 3^c		9^c	10^c		
Europine	10.0–500.0	$y = 4522x - 55375$ 0.999	99 ± 7^a	86 ± 15	7^a	18^a	2.9	9.5
			88 ± 2^b		6^b	20^b		
			70 ± 3^c		5^c	7^c		
Lycopsamine	10.0–500.0	$y = 5103x - 56892$ 0.999	63 ± 4^a	73 ± 8	10^a	16^a	2.5	8.3
			77 ± 1^b		10^b	13^b		
			78 ± 1^c		3^c	9^c		
Europine <i>N</i> -oxide	10.0–500.0	$y = 10367x - 84394$ 0.999	96 ± 1^a	84 ± 12	12^a	20^a	1.4	4.6
			82 ± 2^b		12^b	15^b		
			73 ± 4^c		8^c	8^c		
Intermedine <i>N</i> -oxide	10.0–500.0	$y = 10213x - 124188$ 0.999	75 ± 6^a	81 ± 7	15^a	17^a	2.7	9.1
			79 ± 4^b		11^b	17^b		
			89 ± 7^c		5^c	7^c		
Lycopsamine <i>N</i> -oxide	10.0–500.0	$y = 6870x - 99410$ 0.999	99 ± 7^a	82 ± 19	13^a	19^a	0.9	3.1
			75 ± 1^b		14^b	18^b		
			72 ± 0^c		5^c	9^c		
Retrorsine	10.0–500.0	$y = 55x - 333$ 0.998	93 ± 7^a	97 ± 4	14^a	20^a	2.7	9.1
			100 ± 5^b		16^b	18^b		
			98 ± 4^c		17^c	20^c		
Retrorsine <i>N</i> -oxide	10.0–500.0	$y = 761x + 19568$ 0.999	98 ± 8^a	100 ± 4	17^a	19^a	1.6	5.4
			97 ± 2^b		18^b	20^b		
			105 ± 6^c		7^c	7^c		
Seneciphylline	10.0–500.0	$y = 1540x - 13245$ 0.999	104 ± 10^a	95 ± 12	16^a	16^a	2.3	7.3
			99 ± 2^b		11^b	17^b		
			81 ± 6^c		14^c	16^c		
Heliotrine	10.0–500.0	$y = 26759x - 150017$ 0.999	92 ± 8^a	92 ± 8	11^a	11^a	2.1	7.0
			100 ± 1^b		4^b	14^b		
			85 ± 4^c		9^c	10^c		
Seneciphylline <i>N</i> -oxide	10.0–500.0	$y = 1078x - 14069$ 0.999	90 ± 6^a	90 ± 10	8^a	17^a	2.8	9.2
			99 ± 1^b		11^b	16^b		
			80 ± 9^c		9^c	10^c		
Heliotrine <i>N</i> -oxide	10.0–500.0	$y = 313x + 5661$ 0.999	95 ± 4^a	98 ± 3	8^a	11^a	0.4	1.2
			97 ± 5^b		17^b	18^b		
			101 ± 12^c		8^c	12^c		

Senecivernine	10.0–500.0	$y = 2612x - 34693$ 0.999	94 ± 10 ^a 106 ± 6 ^b 68 ± 7 ^c	89 ± 19	18 ^a 10 ^b 9 ^c	19 ^a 18 ^b 13 ^c	2.2	7.4
Senecionine	10.0–500.0	$y = 2534x - 21778$ 0.999	100 ± 3 ^a 102 ± 0 ^b 76 ± 0 ^c	93 ± 14	15 ^a 13 ^b 9 ^c	16 ^a 16 ^b 14 ^c	1.9	6.3
Senecivernine <i>N</i> -oxide	10.0–500.0	$y = 1098x + 375$ 0.999	96 ± 3 ^a 96 ± 2 ^b 73 ± 7 ^c	88 ± 13	17 ^a 5 ^b 14 ^c	20 ^a 14 ^b 15 ^c	1.9	6.5
Senecionine <i>N</i> -oxide	10.0–500.0	$y = 8706x - 103321$ 0.999	93 ± 5 ^a 96 ± 6 ^b 65 ± 7 ^c	85 ± 17	3 ^a 8 ^b 6 ^c	10 ^a 13 ^b 10 ^c	2.5	8.3
Echimidine	10.0–500.0	$y = 456x - 3313$ 0.999	120 ± 1 ^a 92 ± 2 ^b 102 ± 4 ^c	105 ± 14	3 ^a 8 ^b 17 ^c	3 ^a 10 ^b 17 ^c	2.6	8.8
Echimidine <i>N</i> -oxide	10.0–500.0	$y = 887x + 26203$ 0.998	98 ± 5 ^a 100 ± 1 ^b 104 ± 7 ^c	101 ± 3	11 ^a 15 ^b 6 ^c	11 ^a 19 ^b 19 ^c	2.5	8.2
Senkirkin	10.0–500.0	$y = 10120x - 79673$ 0.999	100 ± 5 ^a 99 ± 4 ^b 85 ± 2 ^c	95 ± 8	19 ^a 10 ^b 5 ^c	20 ^a 17 ^b 12 ^c	1.8	6.0
Lasiocarpine	10.0–500.0	$y = 267x - 3099$ 0.999	111 ± 5 ^a 99 ± 2 ^b 98 ± 2 ^c	103 ± 7	15 ^a 13 ^b 15 ^c	19 ^a 18 ^b 18 ^c	2.3	7.6
Lasiocarpine <i>N</i> -oxide	10.0–500.0	$y = 31643x - 231921$ 0.999	100 ± 6 ^a 91 ± 1 ^b 75 ± 6 ^c	89 ± 13	8 ^a 4 ^b 2 ^c	9 ^a 12 ^b 8 ^c	2.8	9.2

Recovery: mean recovery obtained from six samples ($n = 6$) spiked with the analytes at a known concentration level, and subjected to the proposed extraction procedure; Repeatability: six consecutive injections ($n = 6$) on the same day of a sample spiked with the analytes at a known concentration level; Reproducibility: three replicates of a sample injected in triplicate throughout three different days ($n = 9$) and spiked with the analytes at a known concentration level; MDL: method detection limit; MQL: method quantification limit; ^a Low spiked level (10 µg/Kg); ^b Medium spiked level (100 µg/Kg); ^c High spiked level (800 µg/Kg).

Fig. 1

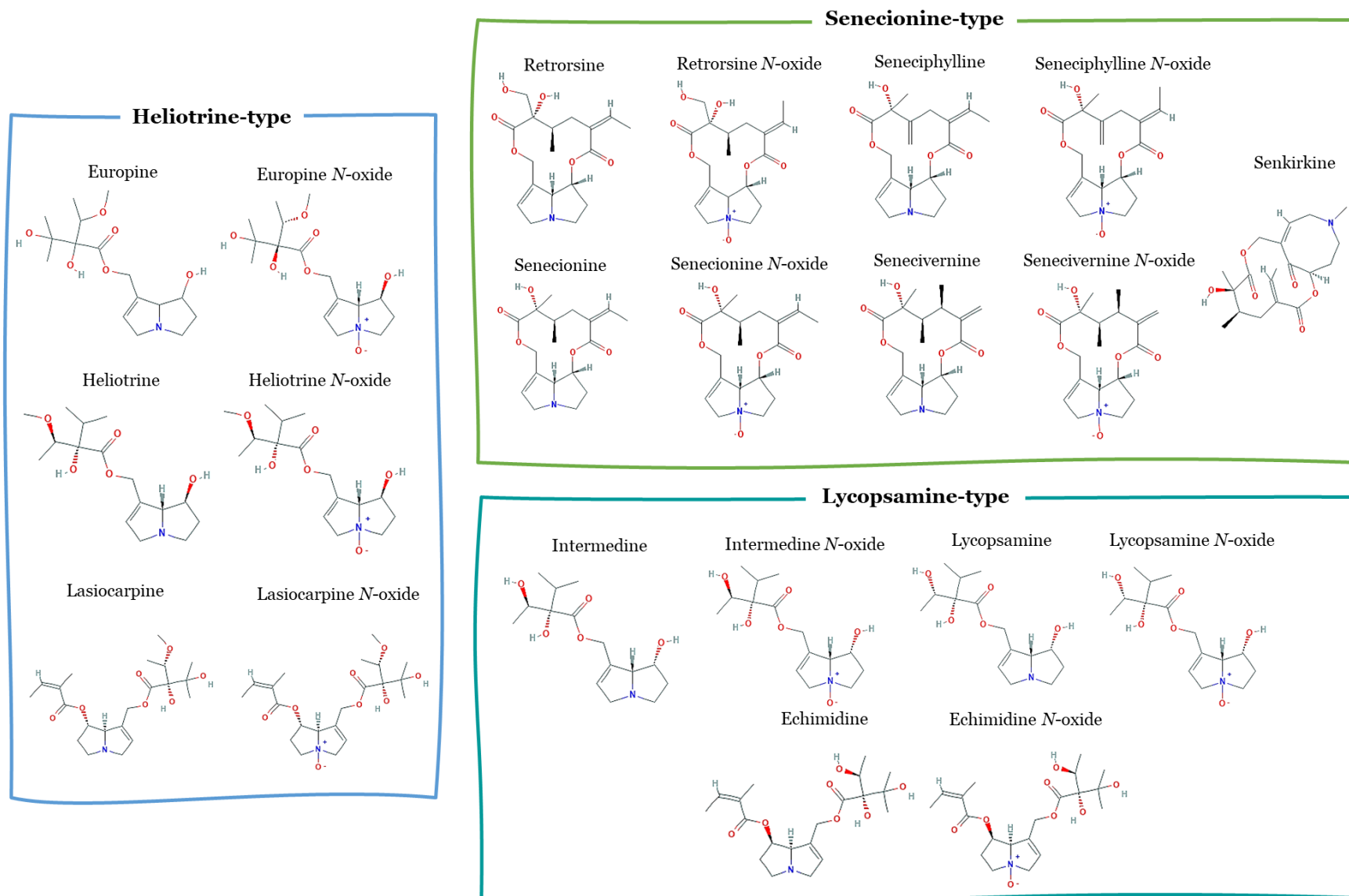


Fig. 2

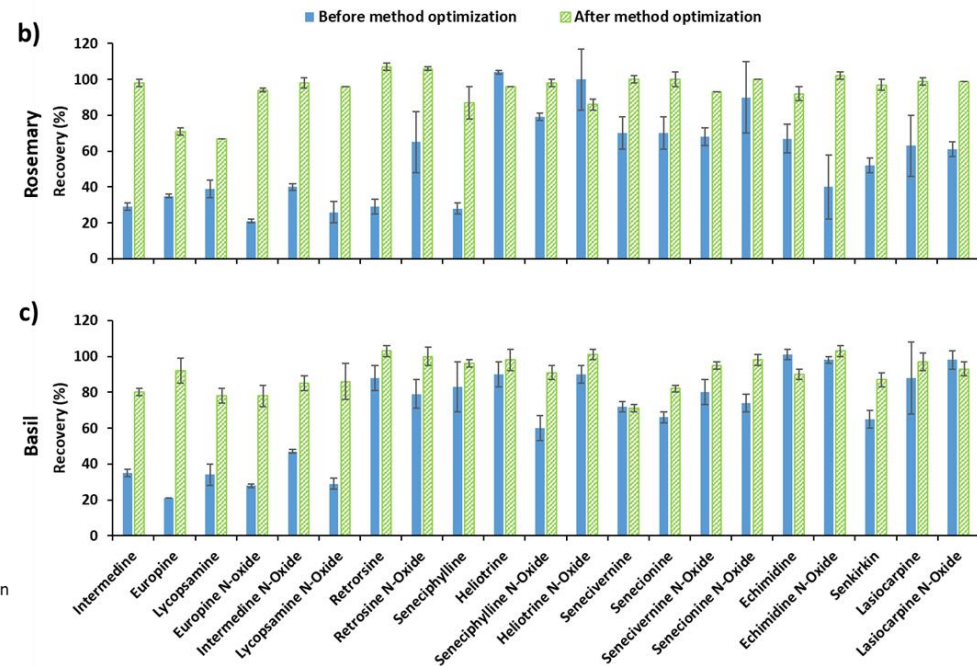
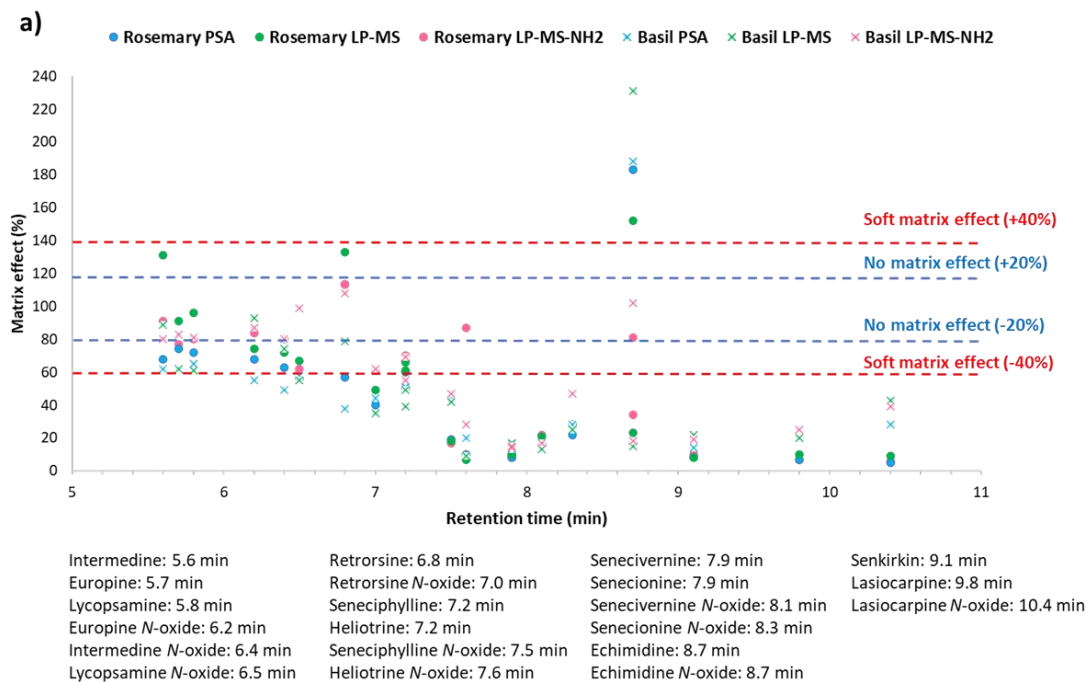
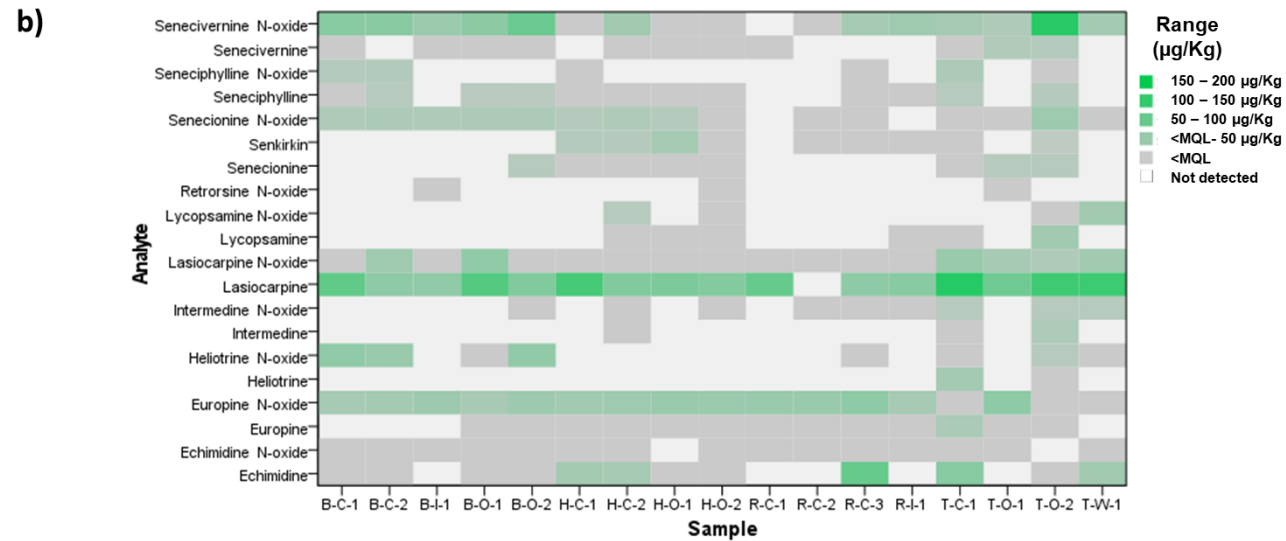
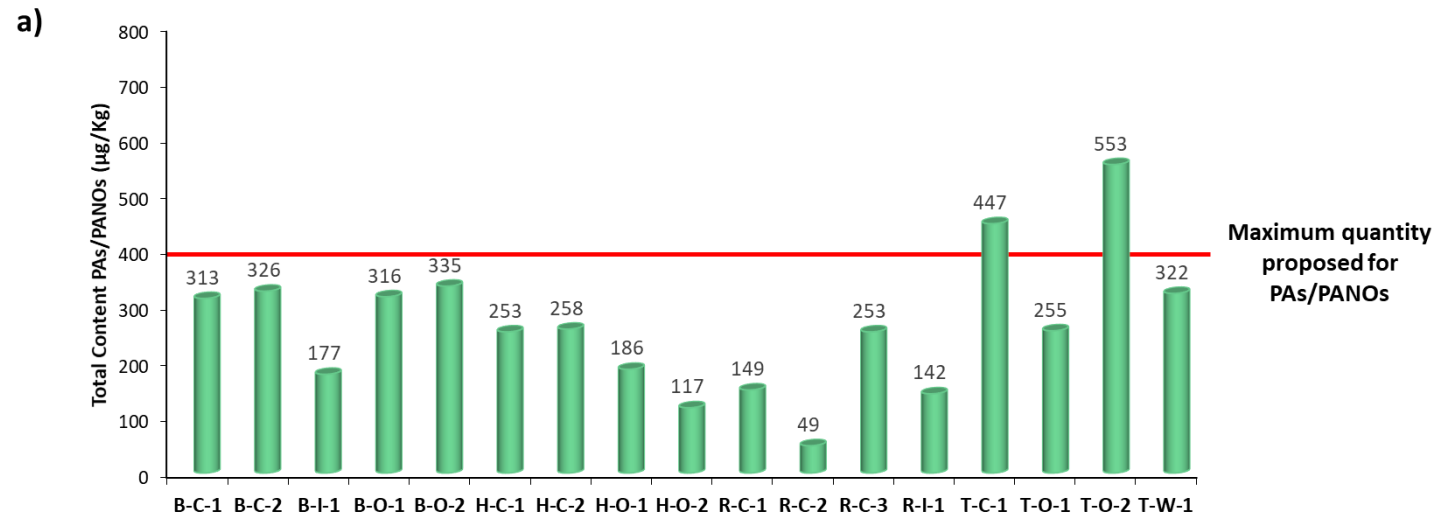


Fig. 3



Supplementary material

Miniaturized and modified QuEChERS method with mesostructured silica as clean-up sorbent for pyrrolizidine alkaloids determination in aromatic herbs

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SM1. Characterization of LP-SM and LP-SM-NH₂ mesostructured silicas

The materials were characterized in terms of transmission electron microscopy (TEM), scanning electron microscopy (SEM) and nitrogen gas adsorption-desorption isotherms. The TEM measures were done using, a JEOL F200 ColdFEG microscope operating at 200 kV with a resolution of 0.23 nm with a microanalysis module EDS JEOL with Centurio detector of 100 mm² and a digital camera One View of GATAN. The samples were dispersed in acetone and placed in a Cu grip with a layer of perforated C. For SEM, a Nova Nano SEM 230 FEI microscope was used. Previously, the samples were prepared with a sputtering method using a sputter coater BAL-TEC SCD 005 as follows: sputter time 100 s, sputter current 30 mA and film thickness 7 nm of gold. Nitrogen gas adsorption-desorption isotherms were carried out on a Micromeritics ASAP 2020 analyzer. Additionally, elemental analysis (% N) was performed on LP-SM-NH₂ to estimate the functionalization degree of amino groups attached to the silica, using a microanalyser model Flash 2000 Thermo Fisher Scientific Inc.

The TEM micrographs of the synthesized materials displayed a perfect pore size distribution in a three-dimensional network, wormhole type. The open-pore network has a homogenous pore size distribution being accessible through the whole particle of the mesostructured silica. SEM micrographs revealed low dispersion particle size with quasi-spherical, spherical, or amorphous shape, with an average size of 100 nm. The particles tend to form clusters or agglomerates with sizes between 1-3 μm common in some silica-based materials. According to the I.U.P.A.C classification, the nitrogen gas adsorption-desorption isotherms of both silicas were of type IV showing an H1 hysteresis loop, which is usual in materials with cylindrical pores of constant cross-section. The isotherms showed several steps: the first step until relative pressure (P/P_0) 0.5 that indicates monolayer adsorption, then at relative pressures between 0.5 and 0.7 takes places the

multi-layer adsorption, between 0.7 and 0.95 the capillary condensation inside the pores of the material and, finally, at higher relative pressures the adsorption on the surface of the material.

Textural properties of both materials are typical of surfactant-assembled mesostructures and verify the uniform framework mesoporosity of the materials synthesized. It was successfully confirmed the large pore volume of the silicas, as the pore volume values obtained (1.74 - 1.18 cm³/g) are higher than the ones for conventional mesostructured silicas. Likewise, the pore size distribution of the materials (calculated with the Barrett-Joyner-Halenda (BJH) method using the desorption branch of the isotherm) was also higher, showing a bimodal pore distribution with the main pore centered at 150 Å , and the second one at 90 Å (in the case of LP-MS). After the modification process, both pore diameters decreased, being higher in the bigger one, as a consequence of a higher functionalization due to its better accessibility and efficient mass-transfer process of the ligand inside-outside the pore in the modification step. The attachment of the amino groups was verified with the % N obtained through elemental analysis, which enabled to estimate the functionalization degree of LP-MS-NH₂

Textural properties of the materials synthesized

Mesostructured silica	S _{BET} ^a (m ² /g)	Pore Volume (cm ³ /g)	BJH ^b pore diameter (Å)	L ₀ (mmol amino groups/g material) ^c
LP-MS	522	1.74	150/90	-
LP-MS-NH ₂	340	1.18	124/89	1.04

^a S_{BET}: Specific surface area calculated by Brunauer-Emmett-Teller (BET) method

^b BJH: Pore size distribution calculated using Barrett-Joyner-Halenda (BJH) method in the desorption branch

^c Functionalization degree of amino groups = (%N × 10)/14

Table S1. Information of the aromatic herb samples analyzed.

Sample	Product origin	Raw material origin	Description
R-C-1	Spain	Unknown	Milled rosemary leaf (<i>Rosmarinus officinalis</i>)
R-C-2	Spain	Unknown	Milled rosemary leaf (<i>Rosmarinus officinalis</i>)
R-C-3	Spain	Unknown	Milled rosemary leaf (<i>Rosmarinus officinalis</i>)
R-I-1	Spain	Spain	Rosemary grown “in-house” (<i>Rosmarinus officinalis</i>)
B-C-1	Spain	Unknown	Milled basil leaf (<i>Ocimum basilicum</i>)
B-C-2	Spain	Egypt	Milled basil leaf (<i>Ocimum basilicum</i>)
B-O-1	France	Unknown	Milled basil leaf from organic farming (<i>Ocimum basilicum</i>)
B-O-2	Spain	Unknown	Milled basil leaf from organic farming (<i>Ocimum basilicum</i>)
B-I-1	Spain	Spain	Basil grown “in-house” (<i>Ocimum basilicum</i>)
T-C-1	Spain	Spain	Milled thyme leaf (<i>Thymus vulgaris</i>)
T-O-1	France	Unknown	Milled thyme leaf from organic farming (<i>Thymus vulgaris</i>)
T-O-2	Spain	Unknown	Milled thyme leaf from organic farming (<i>Thymus vulgaris</i>)
T-W-1	Spain	Spain	Wild thyme on branch (<i>Thymus vulgaris</i>)
H-C-1	Spain	Unknown	Milled herbs de Provence (<i>Satureja hortensis</i> , <i>Rosmarinus officinalis</i> , <i>Ocimum basilicum</i> , <i>Origanum vulgare</i>)
H-C-2	France	Unknown	Milled herbs de Provence (<i>Satureja hortensis</i> , <i>Rosmarinus officinalis</i> , <i>Ocimum basilicum</i> , <i>Origanum vulgare</i> , <i>Origanum majorana</i>)
H-O-1	France	Unknown	Milled herbs de Provence (<i>Satureja hortensis</i> , <i>Rosmarinus officinalis</i> , <i>Ocimum basilicum</i> , <i>Origanum vulgare</i> , <i>Thymus vulgaris</i>) from organic farming
H-O-2	Spain	Unknown	Milled herbs de Provence (<i>Satureja hortensis</i> , <i>Rosmarinus officinalis</i> , <i>Origanum vulgare</i> , <i>Thymus vulgaris</i> , <i>Hyssopus officinalis</i> , <i>Origanum majorana</i>) from organic farming

Table S2. Analytical parameters of the modified μ -QuEChERS method proposed for the determination of the target PAs/PANOs in herbs de Provence samples

Analytes	Linear Range ($\mu\text{g/kg}$)	Matrix-Matched Calibration, R^2	Recovery (% \pm sd)	MDL ($\mu\text{g/kg}$)	MQL ($\mu\text{g/kg}$)
Intermedine	10.0–500.0	$y = 9446x - 94096$ 0.999	89 ± 2	2.9	9.8
Europine	10.0–500.0	$y = 4700x - 34849$ 0.999	88 ± 3	2.4	7.8
Lycopsamine	10.0–500.0	$y = 6258x - 105617$ 0.999	99 ± 1	2.0	6.8
Europine <i>N</i> -oxide	10.0–500.0	$y = 12129x - 131592$ 0.999	96 ± 1	2.7	8.9
Intermedine <i>N</i> -oxide	10.0–500.0	$y = 10315x - 109295$ 0.999	98 ± 3	2.8	9.2
Lycopsamine <i>N</i> -oxide	10.0–500.0	$y = 7669x - 113615$ 0.999	99 ± 3	2.4	8.1
Retrorsine	10.0–500.0	$y = 94x + 1966$ 0.999	96 ± 7	2.5	8.2
Retrorsine <i>N</i> -oxide	10.0–500.0	$y = 1106x - 44992$ 0.999	98 ± 3	3.0	10.0
Seneciphylline	10.0–500.0	$y = 2529x - 42762$ 0.999	97 ± 1	1.9	6.4
Heliotrine	10.0–500.0	$y = 30318x - 43353$ 0.999	82 ± 1	2.5	8.5
Seneciphylline <i>N</i> -oxide	10.0–500.0	$y = 2447x + 77738$ 0.999	100 ± 4	2.9	9.6
Heliotrine <i>N</i> -oxide	10.0–500.0	$y = 338x + 2678$ 0.999	96 ± 1	1.9	6.5
Senecivernine	10.0–500.0	$y = 2391x - 27615$ 0.999	79 ± 5	2.7	9.1
Senecionine	10.0–500.0	$y = 2188x - 24085$ 0.999	80 ± 5	2.0	6.5
Senecivernine <i>N</i> -oxide	10.0–500.0	$y = 1586x + 4472$ 0.999	97 ± 4	2.5	8.2
Senecionine <i>N</i> -oxide	10.0–500.0	$y = 8466x - 134524$ 0.999	95 ± 1	2.1	7.0
Echimidine	10.0–500.0	$y = 153x - 1202$ 0.999	103 ± 2	2.2	7.3

Echimidine <i>N</i> -oxide	10.0–500.0	$y = 2673x + 208480$ 0.999	101 ± 3	1.8	5.9
Senkirkin	10.0–500.0	$y = 2323x - 28876$ 0.999	100 ± 1	2.4	7.9
Lasiocarpine	10.0–500.0	$y = 869x + 78$ 0.999	90 ± 5	2.8	9.2
Lasiocarpine <i>N</i> -oxide	10.0–250.0	$y = 7008x - 80060$ 0.999	94 ± 4	2.8	9.2

Recovery: mean recovery obtained from six samples ($n = 6$) spiked with the analytes at a known concentration level (100 $\mu\text{g}/\text{kg}$), and subjected to the proposed extraction procedure; MDL: method detection limit; MQL: method quantification limit.

Table S3. Solvent-based calibrations (R^2) and matrix effects (ME) calculated for the target compounds in rosemary and basil samples.

Analyte	Solvent-based calibration (R^2)	Rosemary			Basil		
		ME ^a (%)	ME ^b (%)	ME ^c (%)	ME ^a (%)	ME ^b (%)	ME ^c (%)
Intermedine	$y = 20502x + 91927$ (0.999)	68	131	91	62	89	80
Europine	$y = 43117x + 42511$ (0.999)	74	91	77	62	62	83
Lycopsamine	$y = 15490x + 54630$ (0.999)	72	96	80	65	61	81
Europine <i>N</i> -oxide	$y = 69157x + 50258$ (0.999)	68	74	84	55	93	87
Intermedine <i>N</i> -oxide	$y = 24117x + 20670$ (0.999)	63	72	80	49	74	80
Lycopsamine <i>N</i> -oxide	$y = 18251x + 142531$ (0.999)	56	67	62	58	55	99
Retrorsine	$y = 719x + 17758$ (0.999)	57	133	113	38	79	108
Retrorsine <i>N</i> -oxide	$y = 4519x + 24279$ (0.999)	40	49	61	44	35	62
Seneciphylline	$y = 25500x + 154863$ (0.999)	50	61	70	51	39	55
Heliotrine	$y = 49653x + 316712$ (0.999)	60	66	67	54	49	69
Seneciphylline <i>N</i> -oxide	$y = 15171x + 66411$ (0.999)	19	18	17	45	42	47
Heliotrine <i>N</i> -oxide	$y = 9074x - 47681$ (0.999)	10	7	87	20	9	28
Senecivernine	$y = 75086x - 186884$ (0.999)	8	10	9	13	17	14
Senecionine	$y = 75221x - 219174$ (0.999)	8	10	8	14	17	15
Senecivernine <i>N</i> -oxide	$y = 20016x + 171210$ (0.999)	16	21	22	17	13	17
Senecionine <i>N</i> -oxide	$y = 17730x + 176096$ (0.999)	22	28	28	28	25	47
Echimidine	$y = 1609x + 10823$ (0.999)	17	23	34	18	15	18
Echimidine <i>N</i> -oxide	$y = 490x + 3204$ (0.999)	183	152	81	188	231	102
Senkirkin	$y = 35592x - 49230$ (0.999)	8	8	10	14	22	19
Lasiocarpine	$y = 9257x + 111871$ (0.999)	7	10	7	25	20	25
Lasiocarpine <i>N</i> -oxide	$y = 170286x - 317715$ (0.999)	5	9	6	28	43	39

^a ME: matrix effects expressed as the ratio between the slopes of matrix-matched calibration curves (using PSA as clean-up sorbent) and solvent-based calibration curves.

^b ME: matrix effects expressed as the ratio between the slopes of matrix-matched calibration curves (using LP-MS as clean-up sorbent) and solvent-based calibration curves.

^c ME: matrix effects expressed as the ratio between the slopes of matrix-matched calibration curves (using LP-MS-NH₂ as clean-up sorbent) and solvent-based calibration curves.

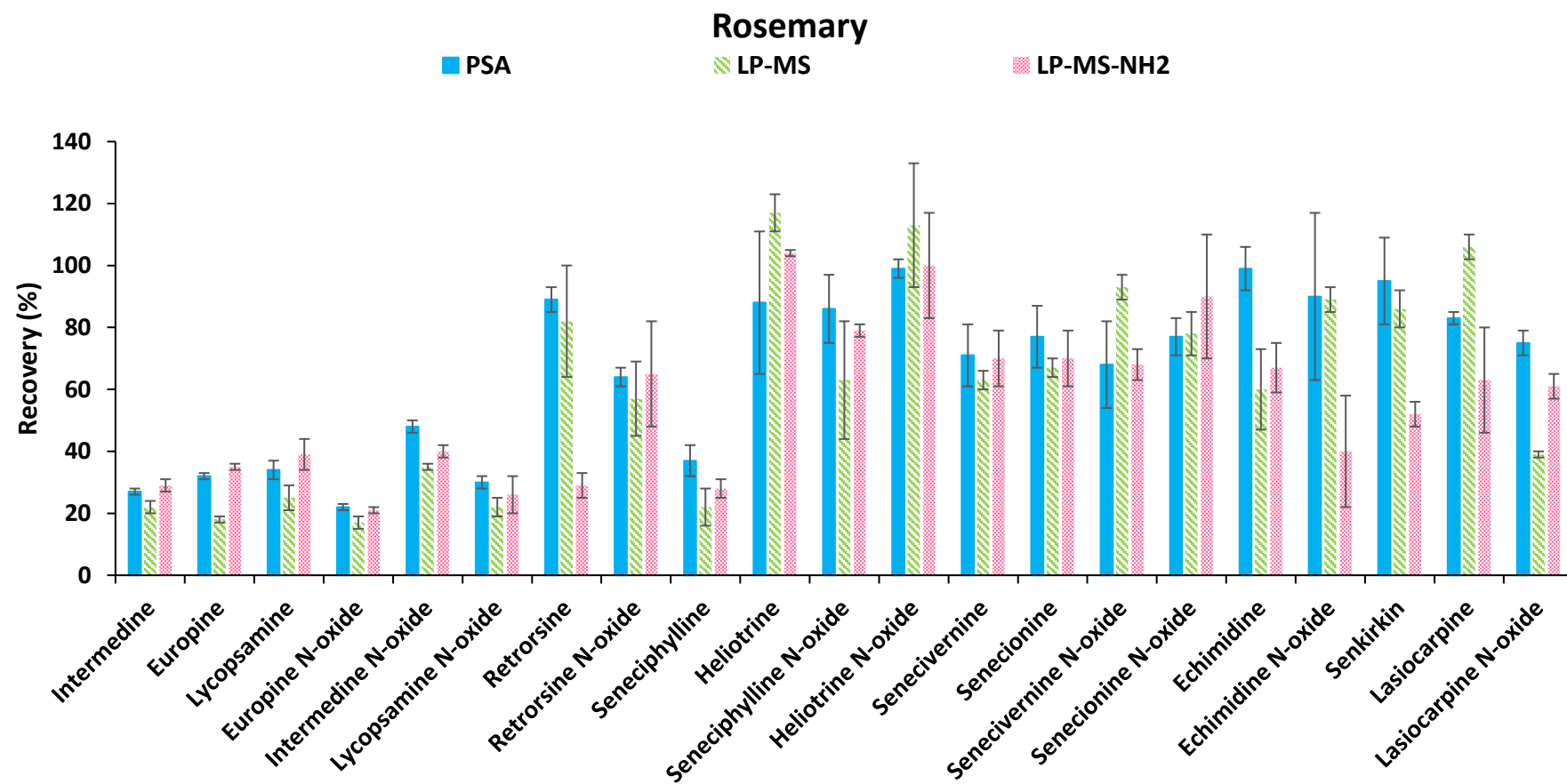


Fig. S1 Comparison of the recovery percentages obtained from the analysis of a spiked rosemary sample (100 µg/kg of each analyte) extracted with the modified µ-QuEChERS procedure proposed before its optimization using PSA, LP-MS and LP-MS-NH₂ as clean-up sorbents. Error bars represent the standard deviation of samples replicates ($n = 3$).

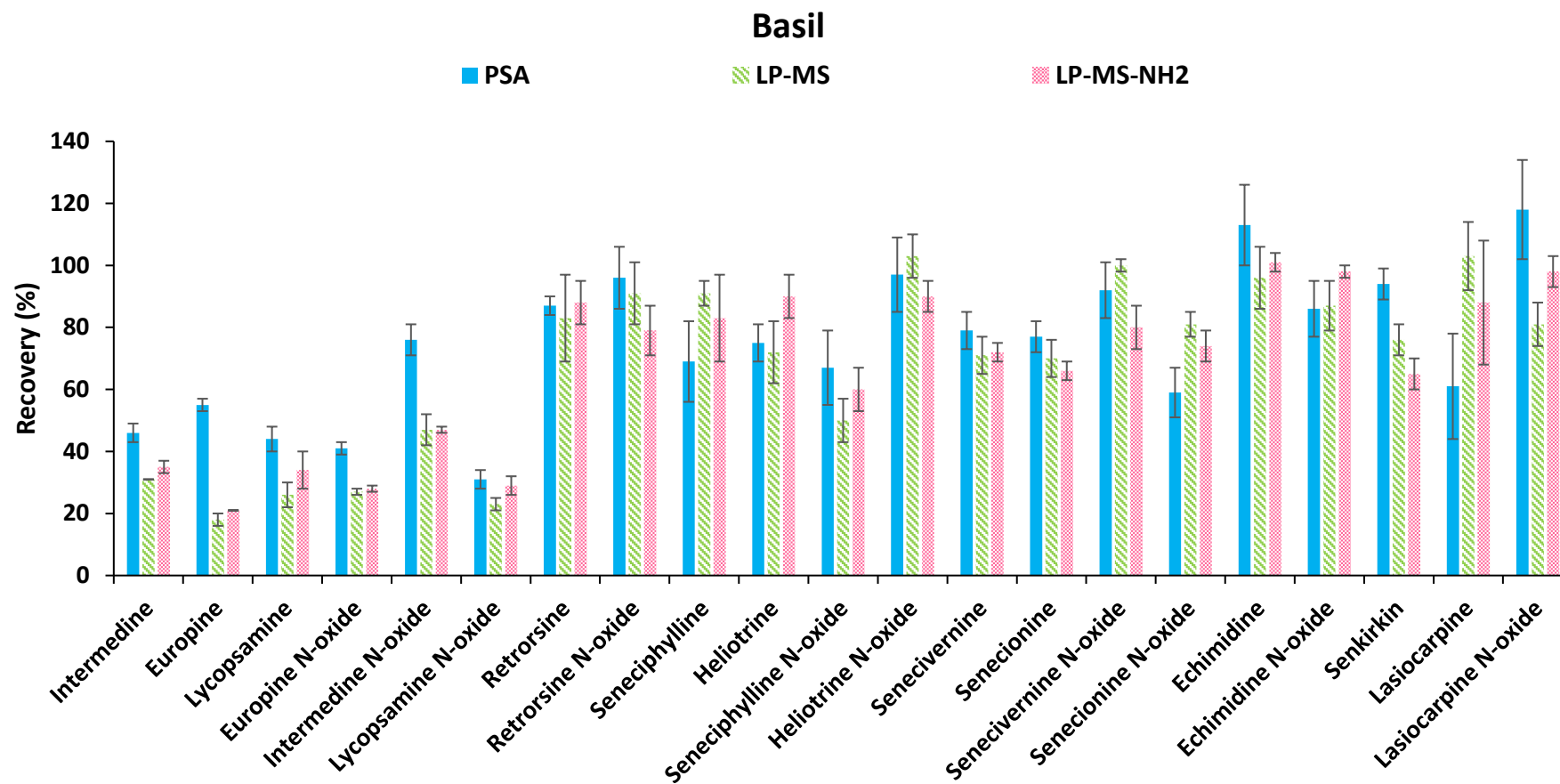


Fig. S2 Comparison of the recovery percentages obtained from the analysis of a spiked basil sample (100 $\mu\text{g}/\text{kg}$ of each analyte) extracted with the modified $\mu\text{-QuEChERS}$ procedure proposed before its optimization using PSA, LP-MS and LP-MS-NH₂ as clean-up sorbents. Error bars represent the standard deviation of samples replicates ($n = 3$).

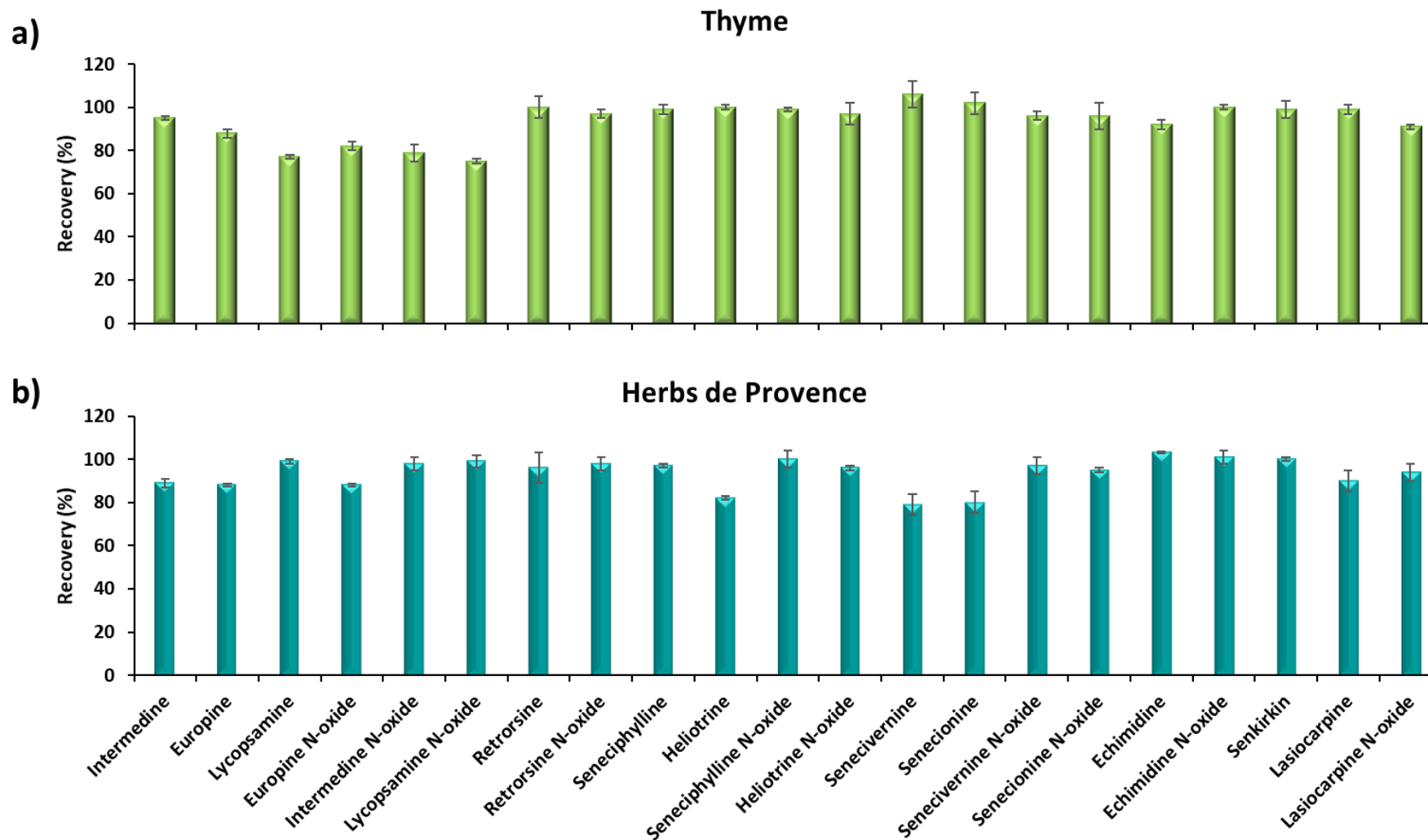


Fig. S3 Recovery percentages obtained from the analysis of (a) three spiked replicates of a thyme sample (100 $\mu\text{g}/\text{kg}$ of each analyte) and (b) three spiked replicates of an herb de Provence sample (100 $\mu\text{g}/\text{kg}$ of each analyte), extracted by the final modified $\mu\text{-QuEChERS}$ procedure proposed. Error bars represent the standard deviation of sample replicates ($n = 3$).

Table S4. Content of the target PAs/PANOs ($\mu\text{g}/\text{Kg}$) quantified in the different aromatic herbs samples analyzed by the modified $\mu\text{-QuEChERS}$ method proposed.

Analytes	R-C-1	R-C-2	R-C-3	R-I-1	B-C-1	B-C-2	B-O-1	B-O-2	B-I-1
Intermedine	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Europine	<MQL	<MQL	<MQL	<MQL	n.d.	n.d.	<MQL	<MQL	n.d.
Lycopsamine	n.d.	n.d.	n.d.	<MQL	n.d.	n.d.	n.d.	n.d.	n.d.
Europine N-oxide	50 ± 1 ^a _d	49 ± 3 ^a _d	59 ± 3 ^b _e	35 ± 2 ^a _a	36 ± 4 ^b _a	38 ± 3 ^c _{a,b}	34 ± 6 ^b _a	47 ± 5 ^c _d	45 ± 2 ^b _{c,d}
Intermedine N-oxide	n.d.	<MQL	<MQL	<MQL	n.d.	n.d.	n.d.	<MQL	n.d.
Lycopsamine N-oxide	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Retrorsine	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Retrorsine N-oxide	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	<LOQ
Seneciphylline	n.d.	n.d.	<MQL	<MQL	<MQL	19 ± 2 ^a _{a,b}	17 ± 5 ^a _{a,b}	16 ± 2 ^a _a	n.d.
Heliotrine	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Seneciphylline N-oxide	n.d.	n.d.	<MQL	n.d.	23 ± 1 ^a _a	24 ± 2 ^a _a	n.d.	n.d.	n.d.
Heliotrine N-oxide	n.d.	n.d.	<MQL	n.d.	57 ± 11 ^c _b	50 ± 17 ^d _b	<MQL	56 ± 20 ^c _b	n.d.
Senecivernine	<MQL	<MQL	n.d.	n.d.	<MQL	n.d.	<MQL	<MQL	<MQL
Senecionine	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	19 ± 3 ^a _a	n.d.
Senecivernine N-oxide	n.d.	<MQL	36 ± 13 ^a _{a,b}	42 ± 10 ^b _{a,b}	65 ± 6 ^d _d	64 ± 5 ^e _d	62 ± 4 ^c _{c,d}	95 ± 17 ^e _e	47 ± 10 ^b _{b,c}
Senecionine N-oxide	n.d.	<MQL	<MQL	n.d.	28 ± 3 ^a _b	30 ± 1 ^b _b	30 ± 1 ^b _b	30 ± 2 ^b _b	28 ± 2 ^a _b
Echimidine	n.d.	n.d.	98 ± 3 ^c _c	n.d.	<MQL	<MQL	<MQL	<MQL	n.d.
Echimidine N-oxide	<MQL	<MQL	<MQL	<MQL	<MQL	<MQL	<MQL	<MQL	<MQL
Senkirkin	n.d.	<MQL	<MQL	<MQL	n.d.	n.d.	n.d.	n.d.	n.d.
Lasiocarpine	99 ± 7 ^b _{c,d,e}	n.d.	60 ± 15 ^b _a	65 ± 7 ^c _a	104 ± 13 ^e _{d,e}	58 ± 2 ^e _a	117 ± 24 ^d _{e,f}	72 ± 1 ^d _{a,b}	57 ± 7 ^c _a
Lasiocarpine N-oxide	<MQL	<MQL	<MQL	<MQL	<MQL	43 ± 3 ^c _c	56 ± 7 ^c _e	<MQL	<MQL
Total	149 ± 35	49 ± 3	253 ± 26	142 ± 16	313 ± 30	326 ± 16	316 ± 36	335 ± 29	177 ± 12

Table S4. (Continued)

Analytes	T-C-1	T-O-1	T-O-2	T-W-1	H-C-1	H-C-2	H-O-1	H-O-2
Intermedine	<LOQ	n.d.	28 ± 6 ^c _a	n.d.	n.d.	<LOQ	n.d.	n.d.
Europine	31 ± 1 ^b _a	<LOQ	<LOQ	n.d.	<LOQ	<LOQ	<LOQ	<LOQ
Lycopsamine	<LOQ	n.d.	41 ± 9 ^d _a	n.d.	n.d.	<LOQ	<LOQ	<LOQ
Europine N-oxide	<LOQ	60 ± 9 ^d _e	<LOQ	<LOQ	40 ± 4 ^b _{a,b,c}	44 ± 1 ^c _{b,c,d}	50 ± 4 ^c _d	47 ± 1 ^a _d
Intermedine N-oxide	19 ± 3 ^a _a	n.d.	20 ± 3 ^{b,c} _a	18.9 ± 0.3 ^a _a	n.d.	<LOQ	n.d.	<LOQ
Lycopsamine N-oxide	n.d.	n.d.	<LOQ	41 ± 2 ^b _b	n.d.	19 ± 2 ^a _a	n.d.	<LOQ
Retrorsine	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
Retrorsine N-oxide	n.d.	<LOQ	n.d.	n.d.	n.d.	n.d.	n.d.	<LOQ
Seneciphylline	19 ± 7 ^a _{a,b}	n.d.	21 ± 3 ^{b,c} _b	n.d.	<LOQ	<LOQ	<LOQ	<LOQ
Heliotrine	37 ± 5 ^b _a	n.d.	<LOQ	n.d.	n.d.	n.d.	n.d.	n.d.
Seneciphylline N-oxide	30 ± 5 ^b _b	n.d.	<LOQ	n.d.	<LOQ	n.d.	n.d.	n.d.
Heliotrine N-oxide	<LOQ	n.d.	20 ± 6 ^{b,c} _a	<LOQ	n.d.	n.d.	n.d.	n.d.
Senecivernine	<LOQ	25 ± 3 ^b _a	23 ± 4 ^{b,c} _a	n.d.	n.d.	<LOQ	<LOQ	<LOQ
Senecionine	<LOQ	19 ± 3 ^a _a	19 ± 5 ^{b,c} _a	n.d.	<LOQ	<LOQ	<LOQ	<LOQ
Senecivernine N-oxide	35 ± 9 ^b _{a,b}	26 ± 3 ^b _a	161 ± 24 ^f _f	39 ± 3 ^b _{a,b}	<LOQ	41 ± 7 ^c _{a,b}	<LOQ	<LOQ
Senecionine N-oxide	<LOQ	<LOQ	44 ± 7 ^d _c	<LOQ	21 ± 1 ^a _a	26 ± 1 ^a _b	22.9 ± 0.2 ^a _a	<LOQ
Echimidine	65 ± 10 ^d _b	n.d.	<LOQ	41 ± 13 ^b _a	38 ± 2 ^b _a	35 ± 5 ^{b,c} _a	<LOQ	<LOQ
Echimidine N-oxide	<LOQ	<LOQ	n.d.	<LOQ	<LOQ	<LOQ	n.d.	<LOQ
Senkirkin	<LOQ	n.d.	11 ± 3 ^{a,b} _a	n.d.	23 ± 3 ^a _b	21 ± 1 ^a _b	36 ± 6 ^b _c	<LOQ
Lasiocarpine	162 ± 13 ^e _h	90 ± 10 ^e _{b,c,d}	139 ± 12 ^e _{f,g}	141 ± 9 ^c _{g,h}	131 ± 16 ^c _{f,g}	72 ± 25 ^d _{a,b}	77 ± 16 ^d _{a,b,c}	70 ± 9 ^b _{a,b}
Lasiocarpine N-oxide	49 ± 3 ^c _d	35 ± 6 ^c _b	26 ± 2 ^c _a	41 ± 2 ^b _c	<LOQ	<LOQ	<LOQ	<LOQ
Total	447 ± 45	255 ± 27	553 ± 48	350 ± 41	253 ± 46	258 ± 18	186 ± 23	117 ± 16

n.d. = not detected; <MQL: below the limit of quantification of the method. In the sample identification code, the first letter indicates: C for conventional farming, O for organic farming and W for wild farming. Different superscript letters in the same column indicate significant differences (p < 0.05) among PAs/PANOs in each sample. Different superscript letters in the same row indicate significant differences (p < 0.05) among samples.

